

## Direct and indirect biomineralization of iron in the ferruginous Lake Pavin

Jasmine Berg<sup>1</sup>, Arnaud Duverger<sup>1</sup>, Didier Jézéquel<sup>2</sup>,  
Elodie Duprat<sup>1</sup>, Fériel Skouri-Panet<sup>1</sup>, Mélanie Poinso<sup>1</sup>,  
Christel Laberty-Robert<sup>3</sup>, Jennyfer Miot<sup>1</sup>

<sup>1</sup>Institut de Minéralogie, Physique des Matériaux et  
Cosmochimie, Sorbonne Universités,

Muséum National d'Histoire Naturelle, CNRS UMR 7590,  
Université Pierre et Marie Curie, IRD 206,

4 Place Jussieu, 75252 Paris Cedex 05, France

<sup>2</sup>Laboratoire de Géochimie des Eaux, Institut de Physique du  
Globe de Paris, Sorbonne Paris Coté, Université Paris Diderot,  
UMR CNRS 7254, 1 Rue Jussieu, 75238 Paris cedex 05, France

<sup>3</sup>Laboratoire de Chimie de la Matière Condensée de Paris,  
UPMC – UMR 7574, Collège de France, 4 Place Jussieu, 75005  
Paris, France

Iron-reducing, iron-oxidizing and sulfate-reducing bacteria are involved in the formation and alteration of Fe-bearing minerals through both direct and indirect biomineralization pathways. The anoxic, iron-rich bottom waters of Lake Pavin are characterized by their unique Fe mineralogy (water-column precipitation of Fe-phosphates) and constitute a perfect habitat for such bacteria potentially involved in Fe cycling. We aimed to investigate the microbial transformations of Fe(III)-phosphate, a key iron phase at the Lake Pavin oxycline, in enrichment cultures with lactate as the electron donor in the presence or absence of added sulfate. Despite low endogenous sulfate concentrations (< 20 µM), sulfate-reducers were the most competitive microorganisms in our enrichments both with and without added sulfate and transformed ferric iron into a diversity of ferrous minerals: vivianite, mackinawite, greigite, and pyrite. As vivianite and pyrite are important mineral phases found *in situ*, we suggest that sulfate and sulfur-reducing bacteria along with iron-reducing bacteria strongly influence iron mineralogy in the water column and sediments of Lake Pavin. Finally, we propose a mechanism for the formation of vivianite ( $\text{Fe}^{\text{II}}_3(\text{PO}_4)_2 \cdot 8\text{H}_2\text{O}$ ) and Fe-sulfides in the water column, whereby amorphous Fe(III)-phosphate is transformed via reductive dissolution or solid-phase rearrangement due to microbial activity.