## Eu isotope variation in Eu standard materials and geological materials

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Europium (Eu) is one of the rare earth elements and the shape of Eu abundance normalized by a chondrite is one of the geochemical proxies to understand an evolutin history of the terrestrial and planetary materials. Eu has two stable isotopes ( $^{151}$ Eu and  $^{153}$ Eu). Generally, it is known that two isotopes do not show isotopic variation within  $\varepsilon$  unit [1,2]. However, recent research results indicate a possibility of slight isotopic variation in Eu isotopes [3,4].

In this work, we measured Eu isotopic ratios precisely from the commercial Eu standard solutions and some reference rock materials. For the standard material, we used a commercial pure Eu solution of NIST3117a. The measurement of Eu isotopes was performed Neptune Plus II MC-ICP-MS at Korea Institute of Geoscience and Mineral Resources. Mass discrimination effect on two europium isotopes was externally corrected by the Sm isotopic composition added to the Eu samples ( $^{150}$ Sm/ $^{154}$ Sm = 0.3244, average value of [5] and [6]) on an exponential law. In our long tests based on the precise measurement and Eu separation technique, we could observe a slight isotopic variation of Eu isotopic ratios in the various Eu standard solutions (0.02%) to 0.06%) as well as references rock materials (-0.67‰ to 0.06‰). This variation indicates a possibility of fractionation processes of europium isotopes in the geological system.

[1] Chang et al. (1994) Int. J. Mass Spectro. Ion Proc. 139, 95-102. [2] Moynier et al. (2006) GCA 70, 4287-4294. [3] Tanaka et al. (2009) Goldschmidt Conf. A1310. [4] Li et a. (2016) Goldschmidt Conf. A1759. [5]Mass and McCulloch (1990) Chem.Geol. 88, 301-315. [6] Hidaka et al. (1995) Anal. Chem. 67, 1437-1441.