

Mineral composition control on inter-mineral iron isotopic fractionation in granitoids

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Previous studies suggested that increase of iron isotope fractionation between high-silica melts and crystallizing minerals is the key to understand the heavier Fe isotopic compositions of high-silica granitic rocks relative to that of dioritic-granodioritic ones. But it is unclear how the β factors of minerals change with the composition variation of their co-existing melts. This study reports elemental and iron isotopic compositions of feldspar and its coexisting minerals from four Dabie I-type granitoids to evaluate the factors that control inter-mineral Fe isotopic fractionation in granitoids. The order of heavy iron isotope enrichment is feldspar > pyrite > magnetite > biotite \approx hornblende. Feldspar has heavier iron isotopic compositions than its co-existing magnetite ($\Delta^{56}\text{Fe}_{\text{plagioclase-magnetite}} = +0.376\text{‰}$ to $+1.084\text{‰}$, $\Delta^{56}\text{Fe}_{\text{alkali-feldspar-magnetite}} = +0.516\text{‰}$ to $+0.846\text{‰}$), which can be attributed to its high $\text{Fe}^{3+}/\text{Fe}_{\text{tot}}$ ratio and low coordination number (tetrahedrally-coordinated) of Fe^{3+} . $\Delta^{56}\text{Fe}_{\text{magnetite-biotite}}$ of coexisting magnetite and biotite ranges from 0.090‰ to 0.246‰ . The large variations of inter-mineral fractionation among feldspar, magnetite and biotite strongly depend on mineral compositions. The $\Delta^{56}\text{Fe}_{\text{plagioclase-magnetite}}$ and $\Delta^{56}\text{Fe}_{\text{alkali-feldspar-magnetite}}$ are positively correlated with albite mode in plagioclase and orthoclase mode in alkali-feldspar, respectively. This could be explained by different Fe-O bond strength in feldspar due to different $\text{Fe}^{3+}/\Sigma \text{Fe}$ or different crystal parameters. The $\Delta^{56}\text{Fe}_{\text{magnetite-biotite}}$ increases with decreasing $\text{Fe}^{3+}/\Sigma \text{Fe}_{\text{biotite}}$ and increasing mole $(\text{Na}+\text{K})/\text{Mg}_{\text{biotite}}$, indicating a decrease of β factor in low $\text{Fe}^{3+}/\Sigma \text{Fe}$ and high $(\text{Na}+\text{K})/\text{Mg}$ biotite.

High-silica leucosomes from Dabie migmatites with a feldspar accumulation petrogenesis have higher $\delta^{56}\text{Fe}$ values ($\delta^{56}\text{Fe} = 0.424\text{‰}$ to 0.567‰) than leucosome that represents pristine partial melt ($\delta^{56}\text{Fe} = 0.117 \pm 0.016\text{‰}$), indicating that accumulation of feldspar could account for high $\delta^{56}\text{Fe}$ values of these rocks. High $\delta^{56}\text{Fe}$ values are also predicted for other igneous rocks that are mainly composed of cumulate feldspar crystals, e.g., anorthosites. Feldspar accumulation, however, cannot explain high $\delta^{56}\text{Fe}$ values of most high-silica granitoids reported in the literature, based on their low Sr, Ba contents and negative Eu anomalies.