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The iron isotope composition of Northern Hemisphere glacial systems

STEVENSON, E.I.*¹, WILLIAMS, H.M.¹, ROBBINS, M.J.², SHEIK, C.S.³, ARENDT, C.A.⁴, CLINGER, A.C.⁵, ACIEGO, S.M.²

¹Department of Earth Sciences, University of Cambridge, Downing St, Cambridge CB2 3EQ. *eis22@cam.ac.uk

²Department of Earth and Environmental Sciences, University of Michigan, Ann Arbor, MI 48109-1005, USA

³University of Minnesota Duluth, Duluth, MN 55812, USA

⁴Los Alamos National Laboratory, Los Alamos, USA

⁵Department of Earth and Planetary Science, University of California, Berkeley, CA 94720-4767, USA

The iron stable isotope composition of subglacial streams draining the Greenland Ice sheet (GIS) have been shown to be highly variable, with dissolved load $\delta^{56}\text{Fe}$ [$(\delta^{56}\text{Fe}, \text{‰} = ({}^{56}\text{Fe}/{}^{54}\text{Fe})_{\text{sample}} / ({}^{56}\text{Fe}/{}^{54}\text{Fe})_{\text{IRMM-14}} - 1) \times 10^3$] ranging from 0 to -2.1 ‰ [1]. Here we expand this data set, by systematically investigating the $\delta^{56}\text{Fe}$ compositions of dissolved loads, suspended sediments, and sediment leachates from glacial and proglacial environments, draining geographically and geologically diverse Northern Hemisphere glaciers. Overall, glaciers draining the Juneau Icefield (JI, Alaska), and the Columbia Icefield (CI, Canada) have fairly invariable $\delta^{56}\text{Fe}$ compositions with both dissolved loads and suspended sediments $\sim 0 \pm 0.2 \text{‰}$.

We utilized leaching techniques that specifically target different Fe-pools within suspended sediments: (i) amorphous Fe (oxy)hydroxides, (ii) crystalline Fe (hydr)oxides, and (iii) residual-Fe, from the GIS, JI and CI sites. Overall, the amorphous (oxy)hydroxides, which represent the most 'potentially bioavailable Fe' [1] are isotopically lighter than both crystalline and residual pools by $\sim 0.1 \text{‰}$. Nevertheless there is overlap between the extracted pools, with all leaches and residual Fe pools falling within $\delta^{56}\text{Fe} \sim 0 \pm 0.2 \text{‰}$.

The experiments above define 'glacial end member' compositions, which may not describe the $\delta^{56}\text{Fe}$ composition and/or flux of Fe that is delivered downstream to the oceans. Therefore we additionally present two short downstream transects of Fe and $\delta^{56}\text{Fe}$ from two outflows draining the south (Qoorqup Sermia) and southeast (Glacier 'G') margins of the GIS, with a late melt season record of $\delta^{56}\text{Fe}$ from the Qoorqup Sermia. Where concurrent measurements have been made, we place the glacial $\delta^{56}\text{Fe}$ compositions in the context of $\delta^{34}\text{S}_{(\text{SO}_4)}$ and $\delta^{18}\text{O}_{(\text{SO}_4)}$ analyses and microbial metabolic function.

[1] Stevenson et al., Gold2016:abs:1718

[2] Bhatia et al., 2013, Nat. Geosci. v. 6, p274-278