

New developments in Li isotope *in situ* analyses with fs-LA-MC-ICP-MS

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Previous studies have shown that metal isotope fractionation at magmatic temperatures is mainly caused by kinetic processes such as diffusion^{1,2}, which can be recorded as isotopic zoning in minerals. The analysis of such diffusion patterns in magmatic olivines and clinopyroxenes of the fast-moving element Li promises insights into the time scales of fast-paced volcanic processes such as cooling and degassing shortly before the eruption of a volcano. *In situ* Li isotope analyses have been performed with SIMS^{3,4} and conventional LA-MC-ICP-MS.⁵ However, measurements by SIMS appear to suffer from composition-dependent matrix effects, and conventional LA-MC-ICP-MS analyses of Li isotope ratios generate relatively large (50-150 μm) and deep laser spots. In order to analyze diffusion profiles of normal-sized magmatic minerals (< 500 μm) a spatial resolution of $\sim 30 \mu\text{m}$ is required. Such a spatial resolution with an ablation depth of only a few μm is achievable using a femtosecond laser ablation system coupled to a Neptune Plus MC-ICP-MS which allows largely matrix-independent metal isotope analyses.^{6,7} Here, we analyzed various silicate reference glasses with low Li concentrations (3-20 ppm) by fs-LA-MC-ICP-MS in order to establish an analytical setup for Li isotope analyses of chemically and isotopically zoned olivine and clinopyroxene crystals with <10 ppm Li.

Materials with Li concentrations down to 3 ppm can be measured with a combination of an ion counter (for ^6Li) and a Faraday cup with a $10^{11} \Omega$ resistor (for ^7Li) with acceptable precisions of 1.2 ‰ (2 RSE). For lower Li concentrations, the combination of ^6Li on an ion counter and ^7Li in a Faraday cup with a $10^{13} \Omega$ resistor seems to be even more promising if a τ -correction is applied to correct for the time delay of the $10^{13} \Omega$ resistor. With this latter combination it should be possible to analyze *in situ* $\delta^7\text{Li}$ values of minerals with Li concentrations of a ~ 1 ppm level precisely.

References

- ¹Jeffcoate et al. (2007); ²Richter et al. (2009);
³Bell et al. (2009); ⁴Su et al. (2015); ⁵Xu et al. (2013);
⁶Oeser et al. (2014); ⁷Lazarov & Horn (2015)