## 'X-Phase' Ca<sub>2</sub>UO<sub>5</sub>•2-3H<sub>2</sub>O: longterm behaviour in natural cementlike rocks

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The analogue of 'X-phase' (Ca2UO5•1.3-1.7H2O) of Atkins et al. [1] has been found as the main secondary phase in the Late Cenozoic marbles with cement-like mineralogy (Tulul Al Hammam natural analogue site, Jordan). The rocks were exposed to supergene alteration during 100 ka that led to the total hydration of primary high-temperature (T=800-850°C) Ca-U(VI) oxides (CaUO4, Ca2UO5, Ca3UO6, Ca4UO7, Ca5UO8, Ca3U2O9 and Ca6UO9) [2]. Natural hydrated Ca-U compound is non-crystalline, more hydrous than 'X-phase' and close to Ca<sub>2</sub>UO<sub>5</sub>•2-3H<sub>2</sub>O (CaO - 21-26; UO<sub>3</sub> - 61-70). Unlike to synthetic 'X-phase' the natural compounds are impured by (in wt%): SiO<sub>2</sub> (0.3-6), F (0.4-4), Al<sub>2</sub>O<sub>3</sub> (0.2-1.8), Fe<sub>2</sub>O<sub>3</sub> (up to 1) and Cl (up to 1.3). It occurs as heterogeneous pseudomorphs after primary Ca-U(VI) oxides and shows specific lamellae-like ('finger') microtexture, typical of synthetic analogue [1]. In some grains naturally-etched tracks of  $\alpha$ -particles are visible. Other secondary U-bearing phases gel-like Ca are silicate hydrates (e.g. (CaO)<sub>3</sub>(UO<sub>3</sub>)<sub>2</sub>(SiO<sub>2</sub>)<sub>2.5</sub>·6H<sub>2</sub>O), while uranophane is very rare.

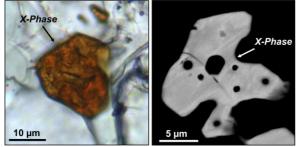


Figure 1: Appearence of natural 'X-phase'.

Thus in natural Ca-rich and high pH environment the U<sup>6+</sup> long-term behaviour is solubility controlled by the formation of hydrous Ca uranates after primary Ca-U(VI) oxides. *Study* was supported by *RFBR grant 15-05-00760*.

Atkins et al. (1988) *Radiochim. Acta* 44/45, 255-261.
Khoury et al. (2015) *Can. Min.* 53, 61-82.