

Interfacial interaction of calcium carbonate in the context of biomedical applications

B. MYSZKA^{1*}, M. SCHÜSSLER², J. HARRIS²,
S.E. WOLF^{2,3}, A.R. BOCCACCINI^{1,3}

¹Institute of Biomaterials, Friedrich-Alexander University
Erlangen-Nürnberg, 91058 Erlangen, Germany
(*correspondence: barbara.myszka@fau.de,
aldo.boccaccini@fau.de)

²Institute of Glass and Ceramics, Friedrich-Alexander
University Erlangen-Nürnberg, 91058 Erlangen,
Germany, (martina.schuessler@fau.de, joe.harris@fau.de,
stephan.e.wolf@fau.de)

³Interdisciplinary Center for Functional Particle Systems
(FPS), Friedrich-Alexander University Erlangen-
Nürnberg, 91058 Erlangen, Germany

Calcium carbonate is one of the most abundant minerals in nature and being widely available and repeatedly proposed for medical applications [1]. The biocompatibility and bioresorbability of CaCO₃ – impressively demonstrated by calcareous biominerals – allows its exploitation as a bioceramic material for implantation purposes, and for other biomedical applications such as drug delivery [2].

Calcium carbonate typically occurs in form of a crystalline polymorph (mainly calcite, aragonite, or vaterite). Also various forms of amorphous calcium carbonate (ACC) exist which can rapidly crystallize at ambient temperature and pressure, especially in a humid environment [3].

In this work, we examine the time-dependent structural evolution of different polymorphs of CaCO₃ (calcite, aragonite, vaterite, ACC) triggered by immersion into different kind of buffer solutions: water, simulation body fluid (SBF) in presence and absence of polymers, respectively. These studies shall provide a profound basis for a future deliberate design of a calcareous bioceramic material.

[1] Addadi *et al.* (2003) *Adv. Mater.* **15**, 12, 959–970.

[2] Lemos & Ferreira (2000) *Mater. Sci. Eng. C*, **11**, 1, 35–

40. [3] Barhoum *et al.* (2015), *J. Mater. Sci.* **50**, 24, 7908–7918.