

Mineral record of abiotic nitrogen reduction

LONG LI¹, VINCENT BUSIGNY², YONG-FEI ZHENG³, KAN LI¹, QIANG HE³, PIERRE CARTIGNY²

¹Department of Earth and Atmospheric Sciences, University of Alberta, Edmonton, Canada T6G 2E3

²Institut de Physique du Globe de Paris, Sorbonne Paris Cité, Univ. Paris-Diderot, 75005 Paris, France

³School of Earth and Space Sciences, University of Science and Technology of China, Hefei 230026, China

Abiotic nitrogen reduction (ANR) in hydrothermal systems has been long considered as the most likely mechanism to transform N₂, the dominant N species on the pre-life Earth's surface, into NH₄⁺, an essential compound for abiotic synthesis of large organic molecules facilitating the emergence of life in the early oceans [1]. Mineral-induced ANR process has been demonstrated by laboratory experiments [2-4], but not yet been verified in natural hydrothermal systems.

To trace the footprint of ANR, we examined NH₄⁺ contents (expressed as N contents below) and N isotope compositions of coexisting muscovite and biotite minerals in 15 granite samples from a fossil hydrothermal system in Eastern China. The samples were formed in a rifting setting of middle Neoproterozoic, where granite H and O isotopes were modified by continental glacial meltwater at high temperature [5]. Our results indicate that muscovite in the samples always have typical granitic signature of low N contents (30-110 ppm) and high δ¹⁵N values (2.5-12.6‰). However, except 2 samples displaying equilibrium NH₄⁺ partitioning and N isotope fractionation between muscovite and biotite, the biotite in all other samples show significant N enrichment (N contents from 110-2400 ppm) coupled with lowered δ¹⁵N values (2.4‰ to -6.8‰). This feature can be only explained by a kinetic isotopic effect [6] associated with N reduction by Fe²⁺ in biotite during fluid-rock interaction. Accordingly, these rocks provide a first geological record for ANR process in natural hydrothermal system.

Reference

[1] Holm, N. & Neubeck, A. (2009) *Geochem. Trans.* 10, doi:10.1186/1467-4866-10-9. [2] Brandes, J.A., et al. (1998) *Nature* 395, 365-367. [3] Smirnov, A., et al. (2008) *Geochem. Trans.* 9, doi:10.1186/1467-4866-1189-1185. [4] Summers, D. P., et al. (2012) *Astrobiology* 12, 107-114. [5] Zheng, Y.-F. et al. (2007) *EPSL* 256, 196-210. [6] Li, L., et al. (2014) *EPSL* 403, 67-78.