

## **Calcite precipitation from CO<sub>2</sub>-H<sub>2</sub>O-Ca(OH)<sub>2</sub> slurry using RESS (Rapid Expansion of Supercritical Solution)**

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The mineral carbonation can sequester CO<sub>2</sub> permanently by converting CO<sub>2</sub> into a thermodynamically stable form of carbonates using Ca or Mg rich raw materials. The direct aqueous mineral carbonation is a method to react CO<sub>2</sub> with Ca or Mg contained raw materials in aqueous solutions to produce carbonates.

In this study, the rapid expansion of supercritical solution (RESS) process was applied to the direct aqueous mineral carbonation to evaluate the feasibility of the RESS process for the direct aqueous mineral carbonation using Ca rich materials. The aqueous mineral carbonation test was conducted on H<sub>2</sub>O-Ca(OH)<sub>2</sub> slurry with supercritical CO<sub>2</sub> at elevated temperature and pressure conditions in a batch reactor to produce micro-sized calcium carbonate. Supercritical CO<sub>2</sub> and Ca solution were reacted at 50 °C, 90 bar and 80 °C, 120 bar for 30 minutes ~ 4 hours, and then reacted Ca slurry was rapidly sprayed through a nozzle to produce calcium carbonate. The crystallized calcium carbonate was analyzed by SEM-EDX and quantitative analysis on polymorphs of calcium carbonate was performed using FT-R and TGA.

The carbonation efficiency was the greatest in the test reacted for 1 hour, regardless of temperature and pressure conditions. At the 50 °C and 90 bar conditions and 1 hour reaction time, the carbonation efficiency was about 93%. In addition, the calcite content was the highest for 1 hour of reaction and the contents of aragonite and vaterite increased after 1 hour of reaction. At the 80 °C and 120 bar conditions, similar results were obtained from the test at the 50 °C and 90 bar conditions, but the changes in the fraction of polymorphs and particle size of crystals were relatively small. These results indicate that carbonate minerals with different polymorphs and sizes can be rapidly produced by controlling temperature, pressure, and reaction time in the direct aqueous mineral carbonation by using the RESS process.