

## Mass-dependent fractionation of strontium isotopes by tree uptake

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Although the measurement of  $^{87}\text{Sr}/^{86}\text{Sr}$  isotopic ratio variations between natural materials has been possible for decades, only recent advances in mass spectrometry have allowed the detection of Sr isotopes mass-dependent fractionation. Only few studies to date have suggested mass-dependent fractionation during plant uptake by comparing  $\delta^{88}\text{Sr}$  between plant tissues and the related bulk soils and rocks underneath [1, 2] or between plant tissues and the exchangeable pools of soils from the same site [3].

We check for any variation of Sr isotopes between rocks, soils and trees on seven Spanish forest stands on contrasted rocks (silicate and calcareous rocks). On each stand, we chose a representative set of samples. We measured the  $^{87}\text{Sr}/^{86}\text{Sr}$  ratio and  $\delta^{88}\text{Sr}$  value of bulk soils and rocks, the corresponding available pools and in the wood of old living trees growing on them. We also compared the  $\delta^{88}\text{Sr}$  in wood from different species of living trees to look for any species-dependent differences. In total 15 pine and 32 oak trees were analysed. Our results show that plants take up their nutrients, and consequently Sr, from the soil exchangeable pool. The  $\delta^{88}\text{Sr}$  values in oaks are always lower than the  $\delta^{88}\text{Sr}$  values of the exchangeable Sr in soils indicating a preferential uptake of the lighter  $^{86}\text{Sr}$  isotope from the available nutrient source. A significant difference in the  $\delta^{88}\text{Sr}$  of pines and oaks is underlined: the higher  $\delta^{88}\text{Sr}$  values in pines compared to oaks suggest a weaker preferential uptake of light Sr by pine species. Experimental studies measuring  $\delta^{88}\text{Sr}$  in rock-soil-plant compartments have to be done in order to identify the processes affecting the Sr isotopes fractionation.

[1] Bullen & Chadwick (2016) *Chem. Geol.* **422**, 25-45.  
[2] de Souza et al. (2010) *Geochim. Cosmochim. Acta* **74**,  
2596–2614. [3] Andrews et al. (2016) *Geochim. Cosmochim.  
Acta* **173**, 284-303.