

Impact of Extreme Pressure on Bonded Radii of Atoms

G.V. GIBBS¹, N.L. ROSS^{1*}, D.F. COX²

¹Department of Geosciences, Virginia Tech, Blacksburg, VA, 24061, USA (*correspondence: nross@vt.edu)

²Department of Chemical Engineering, Virginia Tech, Blacksburg, VA 24061, USA

In order to understand the geochemistry of oxides at the extreme pressures encountered in the Earth's interior, a knowledge of how bonded radii of atoms respond to pressure is needed. Herein, we present results from theoretical calculations of how the bonded radii of Si and O in the SiO₂ polymorphs respond to pressures up to 100 GPa. The Si and O atoms are found to be highly rigid and incompressible. In stishovite, for example, Si and O decrease by 0.05 Å and 0.08 Å, respectively, between 0 and 100 GPa. The impact of extreme pressure on the bonded radii of Al, Ca, La, Sn, and Y atoms in a series of aluminate perovskites has also been examined at high pressure. Each study verifies how little the bonded radii change with pressure. In YAlO₃ perovskite, for example, the Y and Al bonded radii decrease by 0.05 Å and 0.11 Å, respectively, between 0 and 80 GPa. The calculations also indicate a change in coordination of Y with O from 9 to 10 at 20 GPa. The impact of these results on our understanding of the crystal chemistry of oxides under extreme pressure conditions will be discussed.