

Behavior of Mo, U, and Tl isotopes during differentiation in the Kilauea Iki system

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Kilauea Iki lava lake formed during a 1959 eruption and represents a unique natural laboratory for studying magma crystallization and differentiation. Samples with MgO contents ranging from ~27 to 2% have been obtained from drill cores produced by the U.S. Geological Survey, and this sample suite has been used extensively to determine whether non-traditional stable isotope systems experience high-T isotope fractionation. Here, we use these samples to investigate the behavior of the Mo, U (²³⁸U/²³⁵U), and Tl isotope systems during differentiation.

Mo and U show similar chemical behavior in the lava lake system. Both experience mild increases in concentration as MgO decreases from 27% to 8%, followed by a steeper increase over the remainder of the MgO range. No systematic resolvable isotopic change is observed for either element. Mean $\delta^{98}\text{Mo}$ (relative to NIST-3134, set to 0‰) and $\delta^{238}\text{U}$ compositions are $-0.220 \pm 0.079\%$ and $-0.293 \pm 0.095\%$ (2σ stdev).

Tl shows nearly identical chemical behavior to Mo and U, but unlike those other elements, Tl shows significant isotopic heterogeneity. Samples with MgO less than 8% show an increase in $\epsilon^{205}\text{Tl}$ values with decreasing MgO, and three high MgO samples have anomalously high $\epsilon^{205}\text{Tl}$ values ($\gg 0.0\%$). We suggest that these high MgO/high $\epsilon^{205}\text{Tl}$ samples reflect the influence of traces of Fe-Mn sediments previously identified in the Hawaiian mantle source [1], whereas differentiation may have driven the increase in $\epsilon^{205}\text{Tl}$ in the low MgO samples.

[1] Nielsen, Rehkamper, Norman, Halliday, and Harrison (2006), *Nature* 439, 314-317.