

## Effects of seawater Ca/SO<sub>4</sub> and Mg on Sr isotope exchange in MOR hydrothermal systems

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Modern mid-ocean ridge (MOR) hydrothermal systems have been extensively studied for decades. The evolution of seawater-derived fluids as they flow through heated MOR rocks is well understood when the fluids start out with modern seawater chemical composition [1]. However, for much of the past 500 million years, seawater has had higher Ca/SO<sub>4</sub> and lower Mg [2]. These differences can have a substantial effect on chemical and isotopic exchange in MOR's, especially as it relates to Ca, Sr, and Sr isotopes in high-T fluids. We use a charge-balance model calibrated to modern MOR systems and checked with reaction-path modeling to calculate how ancient MOR systems would change due to differences in paleoseawater composition. In modern seawater Ca, Sr, and Ca/SO<sub>4</sub> are low (10 and 0.09 mmol, 0.35) and Mg is high (53 mmol). In mid-Cretaceous seawater, for example, Ca, Sr, and Ca/SO<sub>4</sub> were high (30 and 0.3 mmol, 3.2) and Mg was lower (30 mmol). These differences imply that Cretaceous vent fluids were higher in Ca, Sr, and <sup>87</sup>Sr/<sup>86</sup>Sr. Lower Mg in seawater may also have slowed down chemical exchange processes. This model can explain the high <sup>87</sup>Sr/<sup>86</sup>Sr in Cretaceous ophiolite epidiosites and epidote-quartz veins [3-6], and is consistent with low <sup>87</sup>Sr/<sup>86</sup>Sr in Jurassic ophiolite epidiosites [7]. The model implies that the efficiency of Sr isotope exchange at MOR varies as a function of seawater Ca/SO<sub>4</sub> and Mg. Consequently, modeling the effect of seafloor exchange versus continental weathering in fitting the seawater <sup>87</sup>Sr/<sup>86</sup>Sr history [6] has additional complexity and may imply that continental weathering fluxes varied more than previously thought [7]. The model also has implications for MOR exchange in low-SO<sub>4</sub> Precambrian oceans.

[1] German & Seyfried (2014) *Hydrothermal Processes. The Oceans and Marine Geochemistry*. [2] Lowenstein et al. (2013) *TOG*, v.8 [3] Bickle & Teagle (1992) *EPSL* 113, 219–237. [4] Turchyn et al. (2013) *GCA* 123, 440–458 [5] Coogan (2009) *GG&G*, v.10 [6] Kawahata et al. (2001) *JGR* 106, 11083–11099 [7] Alexander et al. (1993) *JGR* 98, 9731–9759 [8] Halverson et al. (2007) *Paleo*<sup>3</sup> 256, 103–129. [7] Antonelli et al. (2017) *Nature Geosci.* (submitted)