

Experimental study on SO₂ photochemistry under reducing atmosphere and S-MIF

YOSHIAKI ENDO¹, YUICHIRO UENO^{1,2},
SHINNOSUKE AOYOMA¹ AND SEBASTIAN O.
DANIELACHE³

¹Dept. of Earth & Planetary Sciences, Tokyo Tech.,
Japan

²Earth-Life Science Institute (ELSI), Tokyo Tech.,
Japan

³Faculty of Science & Technology, Sophia
University, Japan

Sulfur mass-independent fractionation (MIF) in the Archean sedimentary rock is a key to trace the chemistry of atmosphere at the time. However, the mechanisms of isotopic fractionation originated from SO₂ photochemistry is not fully understood. We conducted photochemical experiments of low $p\text{SO}_2$ (1-10 Pa) condition under 0.1 atm CO atmosphere. The SO₂ column density was down to 10¹⁶ molecules/cm², which should be a realistic SO₂ optical thickness for atmospheric thin conditions as well as providing a reducing condition. Monitoring the gas composition in the chamber together with numerical modeling of the photochemistry revealed complicated reaction pathways in the chamber. Measured isotopic fractionations depend largely on SO₂ column density. Photoproducts (OCS) showed positive $\Delta^{33}\text{S}$, whereas residual SO₂ showed negative $\Delta^{33}\text{S}$. The $\Delta^{36}\text{S}/\Delta^{33}\text{S}$ ratio showed about -1 when ultraviolet spectrum is similar to the solar spectrum (i.e. low $p\text{SO}_2$ condition), that is consistent with the geological record. The $\Delta^{36}\text{S}/\Delta^{33}\text{S}$ ratio reflects combination of the two MIF effects. One came from the self-shielding of SO₂ direct photolysis resulting in the SO₂ column density dependence with $\Delta^{36}\text{S}/\Delta^{33}\text{S}$ slope of ~ -2.4 . Another MIF is derived from photoexcitation of SO₂ showing $\Delta^{36}\text{S}/\Delta^{33}\text{S}$ ratio of $\sim +0.7$ due to inter system crossing (ISC) from singlet SO₂ to triplet SO₂. To transfer the ISC signal into photoproducts, excited state of SO₂ needs to react with reducing gases (e.g., H₂, CH₄ and CO) and to be reduced into SO. Thus, the Archean S-MIF ($\Delta^{36}\text{S}/\Delta^{33}\text{S} = \sim 1$) requires reducing gases in the atmosphere. For example, the late Archean atmosphere ($\Delta^{36}\text{S}/\Delta^{33}\text{S} = \sim -1$; $\Delta^{33}\text{S} > +5$ ‰) may contain high concentration of SO₂ gas (more than 5 ppbv) and of reducing gases (CH₄ or CO) where the ISC-derived MIF contribute 2.3%. Therefore large S-MIF is useful for monitoring partial pressure of SO₂ and reducing gasses.