

Precipitation of Fe-sulfate and Fe-hydroxide intergrowths in laboratory simulations of acid mine drainage: How does co-precipitation affect microtextures?

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In acid mine drainage systems, dissolved iron and sulfate groups in circumneutral to low-pH fluids can precipitate a variety of mineral phases, including ferrihydrite, schwertmanite, goethite, and – in lower pH systems - jarosite, copiapite, melanterite, coquimbite, halotrichite, and rhomboclase. Under specific fluid conditions, these phases can co-precipitate, forming complex intergrowths. Understanding the mechanisms by which these precipitates form is an important step in efforts to prevent and remediate such deposits in acid mine drainage-affected streams. Additionally, efforts to recover and reuse iron from acid mine drainage deposits benefit from an understanding of the microscale structures of these iron-bearing minerals.

Here, we report controlled co-precipitation of Fe-sulfate intergrowths in laboratory-simulated acid mine drainage conditions. Mixtures of 2- and 6-line ferrihydrite $[(\text{Fe}^{3+})_2\text{O}_3 \cdot 0.5\text{H}_2\text{O}]$, goethite $[\alpha\text{-FeO}(\text{OH})]$, schwertmannite $[\text{Fe}^{3+}_{16}\text{O}_{16}(\text{OH},\text{SO}_4)_{12} \cdot 10\text{-}12\text{H}_2\text{O}]$, szomolnokite $[\text{Fe}^{2+}\text{SO}_4 \cdot (\text{H}_2\text{O})]$, potassium jarosite $[\text{KFe}^{3+}_3(\text{SO}_4)_2(\text{OH})_6]$, melanterite $[\text{FeSO}_4 \cdot 7\text{H}_2\text{O}]$, copiapite $[\text{Fe}^{2+}\text{Fe}^{3+}_4(\text{SO}_4)_6(\text{OH})_2 \cdot 20(\text{H}_2\text{O})]$, ferricopiapite $[\text{Fe}^{3+}_{0.67}\text{Fe}^{3+}_4(\text{SO}_4)_6(\text{OH})_2 \cdot 20\text{H}_2\text{O}]$, rhomboclase $[\text{H}_3\text{Fe}^{3+}\text{O}_2(\text{SO}_4)_2 \cdot 2(\text{H}_2\text{O})]$, coquimbite $[\text{Fe}^{3+}_2(\text{SO}_4)_3 \cdot 9\text{H}_2\text{O}]$, and halotrichite $[\text{FeAl}_2(\text{SO}_4)_4 \cdot 22\text{H}_2\text{O}]$ are synthesized, extracted, and subsequently analyzed with X-ray diffractometry and visible- to near-infrared spectroscopy.

Laboratory intergrowths are then compared to naturally-precipitated samples from two regions: 1) circum-neutral acid mine drainage systems issuing from abandoned coal mines in the southern anthracite fields of southeastern Pennsylvania; and 2) low-pH acid mine drainage systems in the Rio Tinto / Rio Odiel drainage basin of the Iberian Pyrite Belt of southern Spain, a site of more than 5000 years of mining-induced acid mine drainage and its related precipitates. Significant intergrowth behavior is reported for both circumneutral and low-pH environments.