

## Reflectance spectroscopy (0.35-2.5 $\mu\text{m}$ ) of low calcium pyroxene + metal mixtures

EDWARD CLOUTIS<sup>1</sup> VISHNU REDDY<sup>2</sup>

<sup>1</sup>Department of Geography, University of Winnipeg,  
515 Portage Avenue, Winnipeg, Manitoba,  
Canada R3B 2E9; (e.cloutis@uwinnipeg.ca)

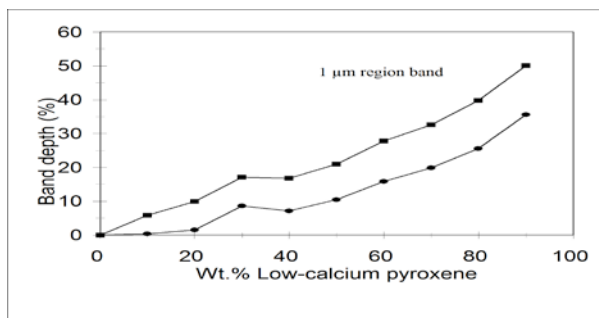
<sup>2</sup>Planetary Science Institute, 1700 East Fort Lowell,  
Suite 106, Tucson, Arizona, USA 85719-2395;  
(reddy@psi.edu)

Low-calcium pyroxene (LCP) and Fe-Ni metal are major components of a number of meteorite classes. There is also spectroscopic evidence for orthopyroxene  $\pm$  metal being the major spectrally active components on a number of asteroids. To improve our ability to determine the composition of such assemblages (pyroxene composition, pyroxene:metal abundance, presence of accessory phases), we undertook a systematic spectral reflectance study (0.35-2.5  $\mu\text{m}$ ) of powdered mixtures of LCP+meteoritic metal produced at 10 wt.% intervals of the end members.

LCP reflectance spectra are characterized by two Fe<sup>2+</sup> absorption bands near 1 and 2  $\mu\text{m}$ , while meteoritic metal is characterized by a featureless, overall red spectral slope.

For the mixtures, increasing amounts of metal cause a number of systematic spectral changes, including: (1) general decrease in reflectance; (2) decrease in the depths of the 1 and 2  $\mu\text{m}$  region pyroxene absorption bands; (3) movement of both LCP absorption band minima to longer wavelengths; and (4) progressive reddening of overall spectral slopes.

The 2  $\mu\text{m}$  band is weaker than the 1  $\mu\text{m}$  band and also shows a more rapid decrease in band depth, becoming largely indistinguishable (<3% deep) at >80 wt.% metal content, while the 1  $\mu\text{m}$  band has a band depth >5% at 90 wt.% metal (Figure 1).



It was found that removal of straight line continua over the 1 and 2  $\mu\text{m}$  absorption bands was sufficient to recover band centers, required for determining LCP composition.