## Durable Technetium Wasteforms and the Challenges They Pose

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Technetium is the lightest element which has no stable isotope. It is of major importance in nuclear medicine (over 80% of all radiopharmaceuticals in the world are labelled with <sup>99m</sup>Tc). <sup>99</sup>Tc is a long life ( $t_{1/2} = 2.1 \times 10^5$  years)  $\beta$ -emitter formed during the fission of U and is a major concern for radioactive waste disposal. The two most common valence states are 4<sup>+</sup> and 7<sup>+</sup>, with 7<sup>+</sup> being known to be highly mobile in the environment. The stability of Tc in waste form materials is of paramount importance for long-term storage. In the last ten years, the development of facilities and techniques appropriate for beta-emitting radionuclides has resulted in the synthesis and analysis of a variety of new Tc-containing oxide materials.

Recently we reported both the structural and magnetic details of the perovskite SrTcO3 and CaTcO3 [1,2]. The issue with this material is that it must be formed at relatively low temperatures and thus it is not very crystalline or dense and so not a suitable wasteform. This presentation will cover recent developments in this area and possible ways forward in wasteform development.

[1]. Rodriguez, E.E., et al., Phys. Rev. Lett., 2011. 106(6).

[2]. Avdeev, M.; et al., J. Am. Chem. Soc. 2011, 133,(6), 1654-1657.