

Past, Present and Future of Mass Independent Isotope Measurements

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The first application of mass independent isotope ratio measurements was nearly 50 years ago when multi sulfur isotope ratio measurements were used to quantify billion year cosmic ray effect spallation effects in iron meteorite based upon the premise that only nuclear effects can produce isotope ratio alterations that are independent mass differences. The subsequent observation of mass independent oxygen isotopic compositions in the primitive Allende calcium inclusion inclusions was interpreted as requiring a nuclear process to account for the mass independent isotopic composition. The Thiemens and Heidenreich ozone experiment in 1983 was the first demonstration that a chemical process was capable of simultaneously producing a mass independent isotopic composition, but one that also duplicates the pattern observed in the calcium aluminum inclusions.

The identification of the quantum mechanical basis of the production of the mass independent isotopic composition remains elusive to this date and in chemical physics has been a subject of intense theoretical and experimental measurements. Likewise, the chemical process responsible for the meteoritic components is a subject of theoretical and physical chemical study to define its role in the early solar system.

In atmospheric molecules it is known that every oxygen bearing molecule has a mass independent, ranging from the second most abundant molecule (O_2) through H_2O , CO_2 , CO , N_2O , H_2O_2 , O_3 , SO_4 , NO_3 , and ClO_4 . This encompasses a range of applications, including global primary productivity quantification present and past (ice core), ozone chemistry and interactions in the troposphere and stratosphere, post Snowball Earth atmospheric chemistry, paleo ozone levels, El Nino global atmospheric chemistry and the record in at the South pole, greenhouse gas source identification and tropospheric chemistry of nitrate and sulfate and the oxygen cycle.

The observation of mass independent sulfur isotopic anomalies in the Archean permits definition of the origin and evolution of oxygen from the oldest rocks through the great oxygenation event and the beginning of defining the total biogeochemical system of sulfur across that time span.