

Effect of iron and trivalent cations on OH-defects in olivine

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Hydrogen incorporation in olivine involves many OH-defects that will control the hydrogen solubility in mantle conditions. Several of these OH-defects are clearly identified essentially from the investigation of the olivine Mg end-member. We study here the effect of Fe²⁺, Fe³⁺, Al³⁺ and Cr³⁺ on OH-defects in order to improve our understanding of the hydrogen speciation in natural olivine. Low-temperature infrared spectra are collected on synthetic and natural olivines, and are interpreted in light of the theoretical determination of the structural, vibrational and infrared spectroscopic properties of Fe-related OH-defects, using first-principles calculations based on density functional theory. The presence of Fe²⁺ changes the cationic environment around the fully protonated vacancies in pure forsterite, leading to a slight modification of their infrared signatures. In particular, the presence of Fe²⁺ in an octahedral site neighbor of the hydrogarnet-type defect is likely responsible for the additional bands observed at 3599 cm⁻¹ and around 3520-3550 cm⁻¹ in Fe-doped olivines. Results show clearly that the OH bands between 3310 and 3380 cm⁻¹ are associated with the presence of trivalent cations. Specifically, two bands at 3323 and 3358 cm⁻¹, commonly observed in natural olivine, are associated with the substitution of Mg²⁺ by Cr³⁺ while two similar bands at 3328 and 3353 cm⁻¹ are associated with the substitution of Mg²⁺ by Fe³⁺. The presence of this defect and the "titanoclinohumite" defect in natural olivine clearly underlines the prominent role of trace elements on the hydrogen incorporation in the lithospheric olivine.