

Microbial CO₂ reduction continuing for millions years in Neogene mudstone resulting in extreme ¹³C-enrichment

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The Horonobe Underground Research Laboratory (URL) project has been conducted for the generic purposes of geoscientific research and R&D on technologies which will be used during the geological disposal. As part of this project, deep boreholes were drilled to obtain information on the hydrogeological characteristics of the Koetoi and Wakkanai formations. Around the boundary between these formations, saline groundwater was diluted with meteoric water [1].

The Horonobe area is on the eastern margin of a Neogene to Quaternary sedimentary basin which is a region of oil and gas accumulation. $\delta^{13}\text{C}$ of paired CH₄ and CO₂ obtained from the project shows relatively heavy values. Previous study [2] reported that CH₄ found in the URL area was affected by microbial secondary oxidation since sedimentation of the Wakkanai Formation. However, another work [3] suggested that CH₄ was produced by microbial CO₂ reduction in a closed system. In this study, we make clear the origin of CH₄.

Our results of $\delta^{13}\text{C} - \delta\text{D}$ systematics are evidence that CO₂ reduction is the primary source of CH₄ in this area. Possible secondary oxidation [2] may be caused in a sample bottle after sampling due to the artifact. Correlation of $\delta\text{D}_{\text{CH}_4-\text{H}_2\text{O}}$ also suggests CO₂ reduction, however, $\delta\text{D}_{\text{CH}_4-\text{H}_2\text{O}}$ around the formation boundary does not match the CO₂ reduction line, which implies that intrusion of meteoric water through fractures would happen relatively recently.

Our results also show that the Koetoi Formation can be assumed an open system regarding to CO₂ supply although the Wakkanai Formation can be a closed system. This is consistent with occurrence of carbonate veins having heavy $\delta^{13}\text{C}$ value mostly in the Wakkanai Formation. In this closed environment, CO₂ reduction continuing for millions years would result in extreme ¹³C-enrichment of CH₄, CO₂, and carbonate.

[1] Ishii *et al.* (2007) *J. Geol. Soc. Jpn* **113** 41-52. [2] Funaki *et al.* (2012) *J. Geograph* **121** 929-945. [3] Tamamura *et al.* (2014) *Chikyukagaku* **48** 39-50.