## Phase relations in the system MgSiO<sub>3</sub>-Al<sub>2</sub>O<sub>3</sub> to 52 GPa and 2000 K

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The chemical composition, structure, mineralogy and dynamics of the Earth's lower mantle can be studied by phase relations in the simplified system in MgSiO<sub>3</sub> - Al<sub>2</sub>O<sub>3</sub> at high pressure and high temperature [1-2]. In this study, phase relations in the system MgSiO<sub>3</sub>-Al<sub>2</sub>O<sub>3</sub> have been determined at pressures of 15-52 GPa and at a temperature of 2000 K using in situ synchrotron X-ray diffraction measurements with sintered diamond anvils in a multi-anvil apparatus. A wide two-phase region of garnet and corundum is found to exist up to 27 GPa, while a phase assemblage of aluminous bridgmanite and corundum is stabilized at higher pressures. The Al2O3 solubility in bridgmanite and the MgSiO3 solubility in corundum are highly dependent on pressure: the Al2O3 content in bridgmanite increases from 12 mol% at 27 GPa to 29 mol% at 52 GPa, forming bridgmanite with the pyrope composition (25 mol% Al<sub>2</sub>O<sub>3</sub>) at about 45 GPa. The new phase diagram in the system MgSiO<sub>3</sub>-Al<sub>2</sub>O<sub>3</sub> confirms that bridgmanite is the dominant host mineral for Al<sub>2</sub>O<sub>3</sub> in the Earth's lower mantle.

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