

A quantitative phase map of nanometric Iron(III) oxide

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There is great interest in nanosized iron oxides, because of their technological application and prospective utilization due to its magnetic property. Among iron oxides, ϵ -Fe₂O₃ is considered as a remarkable phase due to a giant coercive field at room temperature, coupled magneto-electric features, a significant ferromagnetic resonance that is a distinctive characteristic in any other metal oxides [1, 2]. We present the first quantitative phase map for ϵ -Fe₂O₃ via γ - ϵ - α pathway, based on X-ray powder diffraction (XRD) and high-resolution transmission electron microscopy (HRTEM). Our recent study shows the ϵ -Fe₂O₃ nano-minerals occur in oxidized basalt.

The nanosized Fe₂O₃ polymorphs are synthesized by a thermal decomposition method between 800 °C to 1100 °C [3]. The method increases the thermal stability of ϵ -Fe₂O₃ through SiO₂ matrix [3]. The polymorphous transformations of Fe₂O₃ are identified by XRD patterns between 850°C and 1000°C. The individual nanoparticles are examined using HRTEM to study their structure, size, and crystal interfaces. The cubic crystal structure of γ -Fe₂O₃, with the average crystal size of ~5 nm is thermally unstable at elevated temperature. The γ -Fe₂O₃ to ϵ -Fe₂O₃ phase transition takes place once the γ -Fe₂O₃ nanoparticles reach a certain critical size between ~10 nm at 1000 °C and ~13 nm at 850 °C. The orthorhombic crystal structure of ϵ -Fe₂O₃ are typically ~50 nm and ranges from ~10 nm to ~200 nm. The shape of ϵ -Fe₂O₃ is elongated along *a*-axis with combinations of sphenoid, pinacoid and pedion forms. The ϵ -Fe₂O₃ nanoparticles have (110) or (-110) twinning with composition plans of {130} and (100) due to pseudo-hexagonal symmetry of the crystal structure. The critical size of ϵ -Fe₂O₃ to α -Fe₂O₃ phase conversion is approximately between ~150 nm at 1000 °C and ~100 nm at 850 °C in amorphous silica matrix. The rhombohedrally centered hexagonal structure of α -Fe₂O₃ is most stable phase of all three polymorphs of α -Fe₂O₃. HRTEM images show crystal interfaces of ϵ -Fe₂O₃ and α -Fe₂O₃ along *c*-axis. It is proposed that the hexagonal packing of ϵ -Fe₂O₃ (001) surface could serve as the substrate / interface for α -Fe₂O₃.

[1] Gich *et al.* (2005) J. Appl. Phys. 98, 044703. [2] Tucek *et al.* (2010) Chem. Mater. 22, 6483-6505. [3] Kelm & Mader (2005) Anorg. Allg. Chem. 631, 2383-2389.