

Pressure- and temperature-induced transformation of natrolite to sequester Cs and Pb

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We report here a pressure- and temperature-driven transformation to sequester environmentally toxic cations such as Cesium (Cs⁺) and Lead (Pb²⁺). Ag-exchanged natrolite and Pb-exchanged natrolite are pressurized in an aqueous CsI solution and pure water as a pressure-transmitting medium (PTM), respectively. Ag-NAT with CsI solution PTM results in the exchange of Ag⁺ by Cs⁺ in the natrolite framework forming Cs-NAT-I and, above 0.5 GPa, its high-pressure polymorph (Cs-NAT-II). During the initial cation exchange, the precipitation of AgI occurs. Additional pressure and heat at 2 GPa and 160 °C transforms Cs-NAT-II to a pollucite-related, highly dense, and water-free triclinic phase. At ambient temperature after pressure release, the Cs remains sequestered in a monoclinic pollucite phase at close to 40 wt% and favorably low Cs leaching rate under back-exchange conditions [1]. In the case of Pb-NAT in pure water PTM, four high-pressure polymorphs of natrolite (Pb-NAT-I, II, II', III) and one reconstructive transformation are through the pressure-induced hydration (PIH). Heating at 200 °C and 4.5 GPa, leads to a denser lawsonite phase with orthorhombic space group which is recovered after pressure release. The recovered lawsonite phase has much stable Pb coordination environment in the natrolite channel compared to that in the original Pb-NAT and its high-pressure polymorphs [2]. The structures of the recovered Cs- and Pb-sequestered phases are confirmed by HAADF-STEM real spacing imaging.

[1] Im *et al.* (2015) *Environ. Sci, Technol* **49**, 513-519. [2] Im *et al.* (2016) *Dalton Trans* **45**, 1622-1630