

## Smelting process evidenced by $\delta^{66}\text{Zn}$

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Ureilites are highly fractionated ultramafic achondrites that also have some primitive characteristic, such as heterogeneous  $\Delta^{17}\text{O}$  values. They are usually considered to be derived from a single parent body (UPB). The chemical composition and texture of ureilites indicate that they could be mantle residues after ~15-30% basaltic melt extraction during UPB mantle melting [1]. However, such high degrees of partial melting would not preserve the  $\Delta^{17}\text{O}$  heterogeneity of ureilites, as documented in the literature [2].

In this study, we report new oxygen and zinc stable isotopic compositions for five monomict ureilites Yamato (Y) 790981, Y 791538, Y 981750, Y 981810 and Asuka (A) 881931. Our new  $\Delta^{17}\text{O}$  values range from  $-0.493 \pm 0.001\text{‰}$  to  $-1.887 \pm 0.023\text{‰}$ , indicating poor homogenization. In order to reconcile evidence of partial melting with poor homogenization, we propose that the smelting process might explain ureilite formation [3]. Smelting might occur at low degrees of partial melting, with melt extraction taking place too rapidly to homogenize the  $\Delta^{17}\text{O}$  signature [4].

Zn isotope analysis of the five ureilites yielded heterogeneous  $\delta^{66}\text{Zn}$  values ranging from  $+0.61 \pm 0.01\text{‰}$  to  $+1.06 \pm 0.02\text{‰}$ . This heavy isotope enrichment may reflect volatilization processes associated with the smelting process during ureilite genesis [5]. To evaluate the effects of such a volatilization process, we modeled the Zn isotope fractionation in ureilites on the basis of the Rayleigh distillation equation, according to [6] when Zn isotope fractionation was explored during the smelting process in the metallurgic industry. A CI type initial composition was considered for the UPB precursor. Based on this assumption, we show that the observed  $\delta^{66}\text{Zn}$  variability in our ureilites match the data obtained using the smelting process model.

- [1] Kita N., et al. (2004) *GCA*, 68, 4213–4235 [2] Clayton R. & Mayeda T. (1988) *GCA*, 52, 1313–1318 [3] Singletary S. & Grove T. (2003) *MAPS*, 38, 95–108 [4] Goodrich C. et al. (2013), *GCA*, 112, 340-373 [5] Moynier et al., (2010) *Chemical Geology*, 276, 374-379 [6] Mattielli N. et al. (2009) *Atmos. Environ.*, 43, 1265–1272.