Fe²⁺-glutamate intercalated montmorillonite used as photoheterogeneous photo Fenton catalyst at neutral pH

ZHUJIAN HUANG¹, PINGXIAO WU^{2*}

¹ College of Natural Resources and Environment, South China Agricultural University, Guangzhou 510642, P. R. China. (zjhuang@scau.edu.cn)

 School of Environment and Energy, South China University of Technology, Guangzhou 510006, PR China. (*correspondence: pppxwu@scut.edu.cn)

Catalytic Activity Comparison

The catalytic performance of the Fe²⁺-glutamate intercalated montmorillonite (G-Fe-Mt) and the comparison sample (Fe²⁺- montmorillonite, Fe-Mt) was evaluated by the degradation of reactive blue 19 (RB19) under sunlight using H_2O_2 as oxidant. The whole reaction set-up was placed in sunlight (23°02 \square N, 113°24 \square E, Guangzhou, China) for a fixed period of daily between 11:30 a.m. - 14:30 p.m.

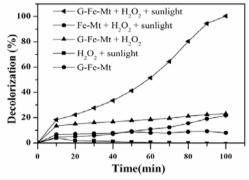


Figure 1: Removal of RB19 in different reaction conditions.

Results and Discussion

It indicated that Fe-Mt has a low reactivity under sunlight irradiation while G-Fe-Mt has excellent ability of light utilization. The photocatalytic reaction of G-Fe-Mt is possibly attributed to electron transfer between glutamic acid organic ligands and irons in chelates. The chelation and the visible light responsiveness of glutamic acid produce a synergistic effect leading to greatly enhanced sunlight-Fenton reaction catalyzed by the heterogeneous G-Fe-Mt under neutral pH [1, 2].

[1] S. Navalon et al. (2010), Appl. Catal. B-Environ. **99**, 1-26. [2] Z. Huang et al. (2014), RSC Adv. **4**, 6500-6507.