

No oxygen added: The influence of temperature on oxygen fugacity in the sub-arc mantle

GLENN A. GAETANI

Dept of Geology and Geophysics, Woods Hole Oceanographic Institution, Woods Hole MA 02543, USA
(correspondence: ggaetani@whoi.edu)

The fugacity of oxygen is a measure of the potential for multi-valent elements – such as iron and carbon – to gain or lose electrons through reaction with oxygen. In magmatic systems, it is typically expressed relative to redox reactions such as fayalite reacting with oxygen to produce magnetite and quartz (ΔFMQ). The ratio of trivalent-to-total iron ($\text{Fe}^{3+}/\Sigma\text{Fe}$) in basaltic glass can be determined either through wet chemistry or X-ray absorption spectroscopy and related to oxygen fugacity using equilibrium thermodynamics, providing proxy for oxygen fugacity of the asthenospheric upper mantle. A comparison of $\text{Fe}^{3+}/\Sigma\text{Fe}$ ratios in ocean floor basalts with those from subduction settings suggests that the relative oxygen fugacity of the sub-arc mantle is significantly higher than that of the oceanic mantle [1]. Results from modeling the behavior of $\text{Fe}^{3+}/\Sigma\text{Fe}$ during adiabatic ascent and partial melting of spinel lherzolite indicate that this difference can be explained by higher mantle temperatures beneath oceanic spreading centers relative to arcs, and that addition of oxygen from the subducted slab is not required.

Modeling involves calculating the $\text{Fe}^{3+}/\Sigma\text{Fe}$ of olivine using the point defect model of [2], and determining $\text{Fe}^{3+}/\Sigma\text{Fe}$ of the bulk peridotite from inter-mineral $\text{Fe}^{3+}/\text{Fe}^{2+}$ exchange coefficients derived from Mössbauer data on natural spinel peridotites, and parameterized in terms of oxygen fugacity, temperature, and the Fe content of the olivine. The $\text{Fe}^{3+}/\Sigma\text{Fe}$ of the melt is determined by combining mass-balance with an equation relating the $\text{Fe}^{3+}/\Sigma\text{Fe}$ of the melt to the fugacity of oxygen [3].

Modeling results indicate that at 2.5 GPa and 1440 °C, a spinel lherzolite with $\text{Fe}^{3+}/\Sigma\text{Fe} = 0.05$ will be at $\Delta\text{FMQ} = -1.95$. A 5% partial melt of this peridotite will have $\text{Fe}^{3+}/\Sigma\text{Fe} = 0.11$. The same peridotite at 2.5 GPa and 1250 °C will be at $\Delta\text{FMQ} = -0.40$, and a 5% partial melt will have $\text{Fe}^{3+}/\Sigma\text{Fe} = 0.19$. Therefore, most – if not all – of the relative oxygen fugacity differences observed between the oceanic and sub-arc mantles reflect temperature.

[1] Kelly & Cottrell (2009), *Science* **325**, 605-607. [2] Dohmen & Chakraborty (2007), *Phys Chem Min* **34**, 409-430. [3] Kress & Carmichael (1991), *Contrib Mineral Petrol* **108**, 82-92.