Electricity in biogeochemistry – potential for confusion?

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Naturally occurring electric fields have been known and investigated in geophysics for decades and have led to discussions of the existence and function of "geobatteries"[1] and "biogeobatteries" [2] where a conducting body catalyzes and transfers electrons between two spatially separated halfreactions. The recently discovered so-called 'cable bacteria' of the Desulfobulbaceae family [3] provide the first example of a fully charactarized natural biogeobattery.

Literature shows that the concepts of electric and redox easily cause confusion. Here potential we present measurements in a cable bacterie microcosm using two types of potentiometric electrodes: so-called redox electrodes and socalled electric potential microelectrodes [4]. The tips of these electrodes are in contact with the environment through exposed platinum and an electrolyte-filled pore, respectively. We will discuss the properties of these electrodes in relation to their responses to the electric and chemical parameters in the system and demonstrate how an interpretation of the data can be used to to reveal details of the localisation and magnitude of the involved electro-chemical processes in the system. Through this, we hope to contribute constructively to the understanding and description of biogeobatteries in terms of electric potentials and redox potentials.

[1] Bigalke & Grabner (1997), *Electrochimica Acta* **42**, 3443-3452. [2] Revil, Mendonca, Atekwana, Kulessa, Hubbard & Bohlen (2010), *Journal of Geophysical Research-Biogeosciences* **115**, G00G02. [3] Pfeffer, Larsen, Song, Dong, Besenbacher, Meyer, Kjeldsen, Schreiber, Gorby, El-Naggar, Leung, Schramm, Risgaard-Petersen & Nielsen (2012), *Nature* **491**, 218-221. [4] Damgaard, Risgaard-Petersen & Nielsen (2014), *Journal of Geophysical Research: Biogeosciences* **119**, 1906–1917.