

Complex microbial ecosystems recorded in sulfur and oxygen isotopes of carbonate associated sulfate in the Monterey Formation, California, U. S. A.

BETHANY P. THEILING¹² AND MAX COLEMAN^{12*}

¹NASA Jet Propulsion Laboratory, California Institute of Technology, Pasadena, California, 91109, U.S.A., max.coleman@jpl.nasa.gov (*presenting author)

²NASA Astrobiology Institute, Wisconsin Astrobiology Research Consortium

$\delta^{34}\text{S}$ and $\delta^{18}\text{O}$ analyses of carbonate associated sulfate (CAS) and mineralogical evidence illustrate a complex microbial ecosystem within the phosphate-rich interval of the Miocene Monterey Formation. All $\delta^{34}\text{S}$ and $\delta^{18}\text{O}$ are below Miocene seawater values ($\sim+22\text{‰}$, VCDT and $\sim+12\text{‰}$, VSMOW). $\delta^{34}\text{S}$ values range from $+12.2\text{‰}$ to $+18.5\text{‰}$ and $\delta^{18}\text{O}$ range from $+2.7\text{‰}$ to $+17.7\text{‰}$, demonstrating a positive linear correlation. $\delta^{34}\text{S}$ of insoluble sediments (combined pyrite and organic sulfur), used as a proxy for bacterially generated H_2S , range from 5.6‰ to 8.4‰ . Sulfate reducing bacteria (SRB) drive the residual seawater sulfate pool to higher $\delta^{34}\text{S}$ values and generate low $\delta^{34}\text{S}$ H_2S ($\sim 14\text{‰}$ lower than seawater sulfate). Initially, H_2S from SRB and Fe^{2+} produced by iron-reducing bacteria (FeRB) generated small amounts of pyrite until all available Fe^{3+} was consumed. With no available Fe^{2+} , H_2S reduced from sulfate by SRB diffused upward to the oxic-suboxic zone boundary where sulfide oxidizing bacteria (SOB) used molecular oxygen to re-oxidize H_2S to sulfate that retains the low $\delta^{34}\text{S}$ of the H_2S . Total sulfate $\delta^{18}\text{O}$ values produced by re-oxidation of H_2S record a mixture of the sulfate pool higher $\delta^{18}\text{O}$ and lower $\delta^{18}\text{O}$ of sulfate from oxidized H_2S . Observed $\delta^{34}\text{S}$ and $\delta^{18}\text{O}$ lower than seawater sulfate illustrate that the $\delta^{34}\text{S}$ versus $\delta^{18}\text{O}$ line defines a mixing line between the endmembers: total reduced sulfur $\delta^{34}\text{S}$ combined with fractionated atmospheric $\delta^{18}\text{O}$ [1], and Miocene seawater sulfate $\delta^{34}\text{S}$ and $\delta^{18}\text{O}$. This interpretation is supported by low CaCO_3 and high phosphate concentrations, whereby SOB promote decreasing seawater pH by production of a small amount of sulfuric acid, which inhibits the precipitation of CaCO_3 while promoting preservation of phosphate [2]. These results demonstrate that the phosphate-rich interval of the Monterey Formation records a complex suite of FeRB, SRB, and SOB, illustrated by $\delta^{34}\text{S}$ and $\delta^{18}\text{O}$ analyses of CAS.

[1] Ziegler (2008) *AGU Fall Meeting Abstracts*, P53D-03 [2] Nathan and Sass (1981) *Chem. Geol.* **34**, 103-111