

First report on Hawaiian carbonatites

ALEXANDER ROCHOLL¹, KLAUS PETER JOCHUM²,
BIRGIT PLESSEN¹, DIETER RHEDE¹, ROLF L. ROMER¹ AND
RICHARD WIRTH¹

¹GFZ German Research Centre for Geosciences,

Telegrafenberg, B121, 14473 Potsdam, Germany

²Max-Planck-Institute für Chemie, Hahn-Meitner-Weg 1,
55128 Mainz, Germany

Carbonatites are common in continental settings but have, so far, only been identified at two oceanic localities, the Cape Verde and Canary Islands, both resting on > 130 Ma old, thick and cool oceanic crust. Here, we report on the first carbonatites observed in a hotter and younger mid-ocean setting, occurring as xenolithic fragments in nephelinitic tuffs at Salt Lake Crater (SLC), Oahu, Hawaii. The existence of Hawaiian carbonatites has been hypothesized before from various lines of evidence: (1) production of carbonatitic melts require carbonate in their source¹; (2) diamond and SiC-bearing SLC xenoliths indicate mixed carbonatitic-silicic metasomatic agents in the sub-Hawaiian mantle² and (3) so do trace element compositions of Niihau³ and SLC⁴ post-erosional magmas.

The potential role of mantle carbonatites on assimilation-fueled ascent of kimberlitic magmas has been the subject of two recent experimental studies^{5,6}. For experimental reasons, both these experiments failed to reproduce conditions prevailing at mantle depth by applying either too low pressure of 1 atm⁵ or using Na₂CO₃ as a substitute for Ca-Mg carbonate^{5,6}. Here, we show the CO₂ producing reaction CaCO₃ + Opx → Cpx + CO₂ “caught in the act” and *in-situ* in sub-Hawaiian carbonatite bearing xenoliths. We document the reaction by means of SEM and TEM analyses and discuss the carbonatites’ isotope composition for C, O and Sr with respect to their parentage.

[1] Brey, *J. Volc. Geoth. Res.* **3**, 61-88, 1978. [2] Wirth & Rocholl, *Earth Planet. Sci. Lett.* **211**, 257-269, 2003. [3] Dixon *et al*, *Geochim. Geophys. Geosyst.* **9**, 1-34, 2008. [4] Rocholl *et al*, *Ann. Meeting German Mineralog. Soc.* (DMG) 2014, **144**. [5] Russel *et al*, *Nature* **481** (2012), 352-356. [6] Kamenetsky and Yaxley, *Geochim. Cosmochim. Acta* **158** (2015), 48-56.