Reducing toxicity of asbestoscontaining waste slate using direct aqueous mineral carbonation

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Direct aqueous mineral carbonation is a method to sequster CO_2 as a form of carbonates with aquous phase. Waste slate is one of the potential alkaline indurstrial wastes for mineral carbonation due to its high content of alkaline earth metals (Ca and Mg). Waste slate primarly contains cement and asbestos material. The asbestos material is considered as a carcinogenic material due to its needle-like morphology.

The feasibility of the direct aqueous mineral carbonation to mitigate the toxicity of asbestos-containing waste slate was evaluation. Leaching and carbonation tests on asbestos-containing waste slate using amonium salts (CH₃COONH₄, NH₄NO₃, and NH₄HSO₄) solutions with various concentrations (0 ~ 4 M) were conducted. The tests were conducted under ambient temperature and pressure conditions.

The Ca leaching efficiency was the highest in CH_3COONH_4 solution (17~35%), whrease the Mg leaching efficiency was the highest in the NH_4HSO_4 solution (7~24%), regardless of a solid dosage. The Ca and Mg leaching efficiencies increased with increasing a solid dosage.The carbonation efficiency for the 1 M CH_3COONH_4 and the 1 M NH_4HSO_4 solutions, which determined using $CaCO_3$ content of the reacted waste slate based on TGA analysis, was approximately 30% for the 1 M CH_3COONH_4 solution and 40% for 1 M NH_4HSO_4 solution. The carbonates were precipitated on the surface of needle-like chrysotile in the reacted waste slate obtained after the carbonation tests. This study results may suggest that the direct aqueous carbonation can mitigate toxicity of needle-like asbestos by changing morphology of asbestos.