

## Generation of condensed carbon from serpentization fluids

V. MILESI<sup>1\*</sup>, F. GUYOT<sup>2</sup>, F. BRUNET<sup>3</sup>, L. RICHARD<sup>4</sup>,  
A. PRINZHOFER<sup>5</sup>, M. BENEDETTI<sup>1</sup> AND T. MCCOLLOM<sup>6</sup>

<sup>1</sup>IPGP, Sorbonne Paris Cité, Univ Paris Diderot, CNRS, F-75005 Paris, France (\*correspondence : milesi@ipgp.fr)

<sup>2</sup>IMPMC, Sorbonne Paris Cité, MNHN, Paris, France

<sup>3</sup>ISTerre, Université Joseph Fourier, Grenoble, France

<sup>4</sup>Carrer de Pontevedra, 08003 Barcelona, Spain

<sup>5</sup>GEO4U, Praia de Botafogo 501, Rio de Janeiro, Brazil

<sup>6</sup>University of Colorado, Boulder, Colorado 80309, USA

Carbonaceous materials were identified in serpentinites bodies from the seafloor and attributed either to an abiotic [1] or a biologic origin [2]. We carried out experimental dissolution of siderite to investigate the reduction of CO<sub>2</sub> in hydrothermal fluids [3]. The produced magnetite was found to be coated by carbonaceous material precipitated from the fluid. Additional experiments with carbon-labeled siderite showed the production of carboxylic acid and methanol at isotopic equilibrium with the siderite.

Compositional paths of serpentization fluids were calculated and compared to natural and experimental fluid compositions. Considering kinetic limitations in the formation of alkanes, calculations showed fluid compositions equilibrating with a carbonaceous material, which buffers the H<sub>2</sub> and CO<sub>2</sub> activities. Fluids from serpentization experiments [4] and from the Lost City [5] and the Rainbow [6] hydrothermal fields are shown to be consistent with this model. Missing amounts of carbon in mass balance extracted from experimental studies are in agreement with the predicted amounts of carbonaceous material, showing that it is the dominant carbon product of CO<sub>2</sub> reduction in these hydrothermal settings.

Formation of carbonaceous material in serpentization fluids would have large implications for the deep Earth carbon cycle by creating an important pool of reduced carbon, and for H<sub>2</sub> production in ultrabasic environments through the intrinsic consumption of H<sub>2</sub>.

[1] Shilobreva et al., 2011; [2] Menez et al., 2012; [3] Milesi et al., 2015; [4] e.g., McCollom and Seewald, 2001, McCollom et al. (in prep.); [5] Proskurowski et al., 2008; [6] Charlou et al., 2002