

Insights into mechanisms of nitrous oxide generation from measurement of nine N₂O isotopologues

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Nitrous oxide (N₂O), a greenhouse gas and ozone-depleting molecule, is currently accumulating in the atmosphere. This accumulation is thought to be driven by biological nitrification and denitrification, but the exact balance of biological contributions of N₂O in the environment is unclear. We are assessing whether distinct isotopic signatures are associated with different biological sources of N₂O, using new analytical methods of high resolution mass spectrometry that enable quantification of: $\delta^{15}\text{N}$, $\delta^{18}\text{O}$, $\Delta^{17}\text{O}$, ¹⁵N site preference, and the clumped isotopologues ¹⁴N¹⁵N¹⁸O, ¹⁵N¹⁴N¹⁸O, ¹⁵N₂¹⁶O, and the sum of ¹⁴N¹⁵N¹⁷O + ¹⁵N¹⁴N¹⁷O. Each of these parameters records different aspects of the substrates and bond-making and bond-breaking reactions that involve N₂O and its precursors, including both equilibrium and kinetic effects.

We have applied these techniques to measure the isotopologues of N₂O associated with a number of environmentally relevant processes, including bacterial and fungal denitrification, bacterial ammonia oxidation, and abiotic nitrite reduction by ferrous iron. The bulk isotopic composition and ¹⁵N site preference for each sample match the expected mechanism of formation and previous pure culture measurements. No sample has position-specific and clumped isotope compositions consistent with an equilibrated final product. Most processes produce a $\Delta(^{14}\text{N}^{15}\text{N}^{18}\text{O} + ^{15}\text{N}^{14}\text{N}^{18}\text{O})$ value, which represents the abundance of these two isotopologues relative to a random distribution of ¹⁵N and ¹⁸O among all isotopologues, between 0 and 1‰. But for some samples that come from organisms with a copper-type nitrite reductase, including both ammonia oxidizing bacteria and denitrifiers, this parameter can be less than zero. One possibility is that these values can be inherited from precursors to N₂O, like nitrite. Finally, the site preference for ¹⁵N in ¹⁸O-containing isotopomers proves to be useful to distinguish N₂O from fungal denitrifiers and the hydroxylamine oxidation pathway of bacterial nitrifiers, which overlap in both $\Delta(^{14}\text{N}^{15}\text{N}^{18}\text{O} + ^{15}\text{N}^{14}\text{N}^{18}\text{O})$ and conventional ¹⁵N site preference. These results suggest that all measured isotopologues are useful for distinguishing among various mechanisms of nitrous oxide production.