

# Theoretical calculation of $^{13}\text{C}$ -D clumped isotope effects in methyl of several organic compounds

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Recent developments in mass spectrometry [1] and tunable infrared laser direct absorption spectroscopy [2] make it possible to measure  $^{13}\text{C}$ -D clumped isotope effects of methane. These techniques can be further applied to determine  $^{13}\text{C}$ -D clumped isotope effects of methyl fragments, therefore need accurate equilibrium  $\Delta_i$  values to calibrate experimental measurements. In this study, we calculate temperature dependences of  $^{13}\text{C}$ -D clumped isotope signatures in methyl of several organic compounds including ethane, propane, acetic acid, etc. Our calculation are performed at CCSD/6-311+G(3df,3pd) by using Gaussian 03 program with no scale treatment.

Our results show that the  $\Delta_i$  values of  $^{13}\text{C}$ -D clumping in methyl fragments of different organic compounds yield similar signals ( $\sim 5.5\text{‰}$  at  $25^\circ\text{C}$ , slightly lower than  $\Delta_i$  value of  $^{13}\text{C}$ -D clumping in methane). For testing the calculated accuracy, theoretical treatments beyond the harmonic level by including several higher-order corrections to the Bigeleisen-Mayer equation are used. Contributions from higher-order corrections (e.g., AnZPE, AnEXC, VrZPE, VrEXC, QmCorr and CenDist) are estimated to repire the ignorings of the Bigeleisen-Mayer equation (the anharmonic effects of vibration, vibration-rotation coupling, quantum mechanics and centrifugal distortion for rotation, etc.) for the calculation of partition function ratios. The results show that the higher-order corrections contribute  $\sim 0.05\text{‰}$  at  $25^\circ\text{C}$ , which is similar to the contribution for calculating  $^{13}\text{C}$ -D clumped isotope signature of methane. By comparing our calculated frequencies to the measured ones, the uncertainty of our calculation of  $\Delta_i$  values  $^{13}\text{C}$ -D clumping in methyl fragments is considered to be within  $\sim 0.05\text{‰}$  at room temperature.

[1] Stolper et al. (2014) *Geochim. Cosmochim. Acta* **126**: 169-191. [2] Ono et al. (2014) *Anal. Chem.* **86**:6487-6494.