

## Tetrahedrally coordinated boron in synthetic high-pressure olenite: Towards Raman spectroscopy as a chemical probe for tourmaline

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Olenitic tourmaline with high amounts of tetrahedral B (up to 2.53 <sup>14</sup>B pfu) has been synthesized in a piston-cylinder press at 4.0 GPa/700 °C and a run duration of 9 days. Crystals are large enough (up to 30 x 150 μm<sup>2</sup>) to allow for reliable and spatially resolved quantification of B by electron microprobe analysis (EMPA), single-crystal X-ray diffraction, and polarized single-crystal Raman spectroscopy. Tourmaline with radial acicular habit is zoned in <sup>14</sup>B-concentration (core: 2.53(25) <sup>14</sup>B pfu; rim: 1.43(15) <sup>14</sup>B pfu) whereas columnar crystals are chemically homogeneous (1.18(15) <sup>14</sup>B pfu). A similar amount of 1.4(1) <sup>14</sup>B pfu was found in the columnar tourmaline by single-crystal structure refinement (SREF) ( $R = 1.94\%$ ). The EMPA identify  ${}^{[T]}\text{Si}_{-1}{}^{[V,W]}\text{O}_{-1}{}^{[T]}\text{B}_1{}^{[V,W]}(\text{OH})_1$  as the main and  ${}^{[X]}\square_{-1}{}^{[T]}\text{Si}_{-1}{}^{[X]}\text{Na}_1{}^{[T]}\text{B}_1$  as minor exchange vectors for <sup>14</sup>B incorporation, which is supported by the SREF. Due to the restricted and well-defined variations in chemistry, Raman bands in the OH-stretching region (3000 - 3800 cm<sup>-1</sup>) are unambiguously assigned to a specific cation arrangement. We found the sum of the relative integrated intensity ( $I_{\text{rel}}$ ) of two low frequency bands at 3284 - 3301 cm<sup>-1</sup> (ν1) and 3367 - 3384 cm<sup>-1</sup> (ν2) to positively correlate with the <sup>14</sup>B concentrations:  ${}^{14}\text{B} [\text{pfu}] = 0.03(1) * [I_{\text{rel}}(\nu1) + I_{\text{rel}}(\nu2)]$ . Hence, those bands correspond to configurations with mixed Si/B occupancy at the T site. Our semi-quantitative correlation also holds for well-characterized natural <sup>14</sup>B-bearing tourmaline from the Koralpe, Austria.

This study contributes to promote Raman spectroscopy as a fast and non-destructive tool for the chemical classification of (precious) natural tourmaline.