

Modelling the behavior of natural U and Ra in Forsmark, Sweden

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It has been proposed that the fractured crystalline bedrock of Forsmark, Sweden host a spent nuclear fuel repository. Early site characterizations found elevated dissolved U in groundwater[1], prompting studies on the local origin and solubility of natural U. At Forsmark, hydrothermal fluid circulation (>1Ga) is connected to U oxidation and alteration of primary (U,Th)-minerals to U(VI)-silicates haiweeite and uranophane[2].

Water-rock interaction with respect to U has been investigated using U/Th-series disequilibrium and PHREEQC geochemical modelling. Eleven groundwater samples with 0.39-150 $\mu\text{g/L}$ dissolved U were measured for $^{224,223}\text{Ra}$ ($\tau_{1/2}<12\text{d}$) and $^{226,228}\text{Ra}$ ($\tau_{1/2}<1600\text{y}$) using a Radium Delayed Coincidence Counter and gamma spectrometry.

Activity ratios (ARs) of U/Th-series nuclides offer insight into the rates of processes (alpha recoil, dissolution, sorption, advection etc.) that control their groundwater distributions upon interaction with a source, i.e. local minerals and the rock surface. Most Forsmark samples are correlated by $^{222}\text{Rn}/^{226}\text{Ra}$ and $^{223}\text{Ra}/^{235}\text{U}$ ARs, indicating relations between the above processes. The sample with the highest dissolved U (150 $\mu\text{g/L}$) has anomalously high ^{235}U activities, which may indicate that a distinct set of processes and/or rates have enhanced the U-series disequilibrium. The sample with second highest dissolved U (34 $\mu\text{g/L}$) has $^{226}\text{Ra}/^{238}\text{U}$ and $^{223}\text{Ra}/^{235}\text{U}$ ARs >2 orders of magnitude higher than the 150 $\mu\text{g/L}$ U sample. Speciation-solubility calculations made using PHREEQC results have suggested that both samples are undersaturated with respect to uranophane and U speciation is dominated by low-sorbing $\text{Ca}_2\text{UO}_2(\text{CO}_3)_3^0$ and $\text{CaUO}_2(\text{CO}_3)_3^{2-}$. These ARs may provide further insight into e.g. respective rates and/or timing U dissolution and adsorption at Forsmark.

[1] Smellie *et al.* (2008) SKB R-08-84. [2] Krall *et al.* (2015) *Applied Geochemistry*.