

# Controls on the molybdenum isotope composition of river water: Insights from Iceland and New Zealand

K. HORAN<sup>1\*</sup>, R. G. HILTON<sup>2</sup> AND K. W. BURTON<sup>1</sup>

<sup>1</sup>Department of Earth Sciences, Durham University, UK

(\*correspondence: kate.horan@durham.ac.uk)

<sup>2</sup>Department of Geography, Durham University, UK

The molybdenum (Mo) isotope composition of river water plays a central role in setting the  $\delta^{98/95}\text{Mo}$  of the oceans [1] [2]. Use of  $\delta^{98/95}\text{Mo}$  as a palæo-redox proxy in marine sediments requires quantification of the controls on  $\delta^{98/95}\text{Mo}$  in continental runoff. Additionally, Mo can be sourced from sulfides and organic carbon, and its mobilisation may help quantify oxidative weathering reactions which govern the long-term evolution of atmospheric  $\text{CO}_2$  and  $\text{O}_2$  [3] [4]. Here we present measurements of redox-sensitive trace elements Mo and rhenium (Re) and  $\delta^{98/95}\text{Mo}$  in rivers with contrasting lithology and differing erosion processes (glacial versus non-glacial) to understand dissolved  $\delta^{98/95}\text{Mo}$  variability.

Existing  $\delta^{98/95}\text{Mo}$  data from glacial rivers in basaltic catchments in Iceland [2] show  $\delta^{98/95}\text{Mo}$  values increase from  $\sim 0\text{‰}$  close to the source to  $\sim 1\text{‰}$  downstream. This is thought to reflect weathering of basalt ( $0\text{‰}$ ) and isotope fractionation downstream associated with oxy-hydroxide formation or cycling of Mo through organic matter. New data from the Skaftá glacial river indicates higher  $\delta^{98/95}\text{Mo}$  values at the source ( $0.43\text{‰}$ ) which decrease downstream ( $0.23\text{‰}$ ), suggesting a role of the rhyolitic basement geology in this catchment and a predominance of sulfide weathering. A sulfidic source of Mo is supported by high sulfate concentrations.  $\delta^{98/95}\text{Mo}$  data from the Skaftá River also support the premise that glacial systems are isotopically heavy compared with non-glacial systems [2] with a non-glacial tributary having  $\delta^{98/95}\text{Mo} = 0.08\text{‰}$ . Sediment with high surface area to volume ratio in glacial catchments may facilitate increased isotopic fractionation.

Preliminary dissolved Re concentrations in rivers draining meta-sedimentary bedrock in the western Southern Alps, New Zealand, suggest that the efficiency of oxidative weathering reactions is much greater in glacial ( $[\text{Re}] = 1.3\text{ppt}-1.5\text{ppt}$ ) compared with non-glacial catchments ( $[\text{Re}] = 0.4-0.7\text{ppt}$ ). Application of  $\delta^{98/95}\text{Mo}$  in these systems will lend new insights into the role of Mo source versus isotope fractionation during weathering in these catchments.

[1] Archer & Vance (2008) *Nature Geosci.* **1**, 597-600. [2] Pearce *et al.* (2010) *EPSL* **295**, 104-114. [3] Calmels *et al.* (2007) *Geology* **35**, 1003-1006. [4] Hilton *et al.* (2014) *EPSL* **403**, 27-36.