

Photochemical and biological lability of soot black carbon in soils from Phoenix, AZ

HILAIRY HARTNETT^{1,2*} AND GEORGE A. HAMILTON²

¹School of Earth and Space Exploration, Arizona State University, Tempe AZ 85287 (*correspondence: h.hartnett@asu.edu)

²Department of Chemistry and Biochemistry, Arizona State University, Tempe AZ 85287

Soot BC is the product of incomplete combustion of biomass and fossil fuels. Soot BC is traditionally thought to be chemically non-reactive because it is highly aromatic and has a low O:C ratio; however, the inherent imbalance between sources and sinks in the global black carbon budget suggests soot BC in soils may be considerably more reactive than previously considered. Soil is a potential storage reservoir for soot BC especially in urban regions with significant fossil fuel combustion. Phoenix, AZ is a rapidly urbanizing arid region with relatively low bulk soil organic carbon content; soot BC is a significant fraction of the soil organic matter and may play a major role in soil biogeochemical cycling in this region (Hamilton and Hartnett, 2013). Soot BC:OC ratios in soils from the CAP-LTER Survey 200 are as high as 30% which is among the highest reported in any soils.

We investigated both photo-chemical oxidation and microbial degradation of soil soot BC. Soot BC was readily photo-oxidized by UV light in the laboratory; we observed a loss of ~19% of the soil soot BC over ~24h. This corresponds to a solar-equivalent degradation rate of 0.55 ± 0.1 g soot BC/kg soil/y. In mesocosm biodegradation experiments CO₂ production by soil microbes increased by 64 to 100% over controls, even when soot BC was the only carbon source present. Both photo-oxidation and biodegradation not only reduce the amount of soot BC present in soils but also alter the chemical composition of soot BC as determined by Fourier-Transform Infra-Red (FTIR) spectroscopy and 2D fluorescence analysis. Taken together, these results strongly imply that soot BC plays an active role in the soil biogeochemical cycling of this urban desert ecosystem.

[1] Hamilton & Hartnett (2013) *Org. Geochem.* **59**, 87-94.