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## Melting the shallow, waterundersaturated mantle wedge at subduction zones

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## Introduction

Mantle melting processes at subduction zones are controlled by the presence of water and affected by inputs from the subducting slab. While much progress has been made in understanding the beginning of mantle melting at the slabmantle wedge interface, relatively little work has been undertaken to investigate the melts that form in the middle and upper sections of the mantle wedge. In this study, we perform forward experiments on two mantle compositions to investigate the roles of pressure, temperature, water, and bulk composition on the composition of mantle melts.

## Methods

Piston cylinder experiments investigate 1) a primitive upper mantle composition at 1.0, 1.2, 1.6, and 2.0 GPa under water-undersaturated conditions (0-6 wt% H<sub>2</sub>O) at 1135-1470 °C and 2) a depleted upper mantle composition at 1.2 GPa with small amounts of water (1-3 wt% H<sub>2</sub>O) at 1225-1275 °C. Both mantle compositions have been modified to include a  $\leq 0.5\%$ slab component in the form of additional Na<sub>2</sub>O and K<sub>2</sub>O.

## Results

Melt compositions from both primitive and depleted experiments confirm trends reported by Gaetani and Grove [1], which indicate that increasing H<sub>2</sub>O at constant melt fraction F and pressure results in increases of SiO2 and Al2O3 and decreases of MgO and FeO. CaO and Na2O remain constant. The results from this study also reveal the role of pressure with constant, small amounts of water at constant F. For primitive experiments conducted at constant F and H2O, decreasing pressure causes a slight increase in SiO<sub>2</sub> and a decrease in FeO, with MgO, Al<sub>2</sub>O<sub>3</sub>, CaO, and Na<sub>2</sub>O remaining constant.

In the primitive experiments, all major elements except for CaO reproduce the composition of natural primitive, high-Mg andesites (PHMA), rocks that are thought to be direct mantle melts. Curiously, the CaO contents of PHMA from various subduction zones are resolvably different and range from 6 wt% to 10 wt% CaO. That this range in CaO is not achieved in our experiments while the range of Al2O3 is reproduced suggests that the range of CaO in the natural samples is caused by heterogeneities in the Ca/Al ratio of the upper mantle.

[1] Gaetani and Grove (1998), CMP 131, 323-346