

Lack of calcium isotope fractionation during magma differentiation

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Recent studies of non-traditional stable isotopes, such as Mg, Ca and Fe, in silicate rocks show measurable isotopic variations: some reflecting isotopic fractionation between different phases, and some being attributed to recycling of surface material into deep mantle. For example, $^{44}\text{Ca}/^{40}\text{Ca}$ in orthopyroxenes are higher than that in the co-existing clinopyroxenes in mantle peridotites, which reflect inter-mineral isotopic fractionation controlled by the strength of Ca–O bond [1]. This observation leads to an important question: Is there Ca isotopic fractionation during magmatic differentiation?

Basalts from Kilauea Iki lava lake, Hawaii, USA were produced by closed-system crystal-melt fractionation, with MgO ranging from 26.87 to 2.37 wt% [2]. They define a negative MgO–CaO trend at MgO > 7.5 wt%, reflecting olivine fractionation/accumulation, and a positive MgO–CaO trend at MgO < 7.5 wt%, controlled by clinopyroxene and plagioclase fractionation. Within analytical errors, they do not show measureable isotopic fractionations for Li and Mg [3, 4]. By contrast, $\delta^{56}\text{Fe}$ and $\delta^{66}\text{Zn}$ increase with decreasing MgO content in Kilauea Iki lavas, implying Fe and Zn isotopic fractionation introduced by crystal fractionation/accumulation [5, 6].

Here we present high-precision mass-dependent Ca isotopic analyses on nine basalts from Kilauea Iki lava lake using a ^{43}Ca – ^{48}Ca double spike technique. Although these samples have undergone clinopyroxene and plagioclase differentiation (up to about 20 wt% for each phase), they show no resolvable variations in Ca isotope composition, with an average $\delta^{44/40}\text{Ca}_{\text{SRM915a}}$ of 0.91 ± 0.09 (2σ).

[1] Huang *et al* (2010) *EPSL* **292**, 337–344. [2] Helz (1987) *Magmatic Processes: Physicochemical Principles*. pp. 241–258. [3] Tomascak *et al* (1999) *GCA* **63**, 907–910. [4] Teng *et al* (2007) *EPSL* **261**, 84–92. [5] Teng *et al* (2008) *Science* **320**, 1620–1622. [6] Chen *et al* (2013) *EPSL* **369–370**, 34–42.