

## Lack of calcium isotope fractionation during magma differentiation

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Recent studies of non-traditional stable isotopes, such as Mg, Ca and Fe, in silicate rocks show measurable isotopic variations: some reflecting isotopic fractionation between different phases, and some being attributed to recycling of surface material into deep mantle. For example, <sup>44</sup>Ca/<sup>40</sup>Ca in orthopyroxenes are higher than that in the co-existing clinopyroxenes in mantle peridotites, which reflect inter-mineral isotopic fractionation controlled by the strength of Ca–O bond [1]. This observation leads to an important question: Is there Ca isotopic fractionation during magmatic differentiation?

Basalts from Kilauea Iki lava lake, Hawaii, USA were produced by closed-system crystal-melt fractionation, with MgO ranging from 26.87 to 2.37 wt% [2]. They define a negative MgO–CaO trend at MgO > 7.5 wt%, reflecting olivine fractionation/accumulation, and a positive MgO–CaO trend at MgO < 7.5 wt%, controlled by clinopyroxene and plagioclase fractionation. Within analytical errors, they do not show measureable isotopic fractionations for Li and Mg [3, 4]. By contrast,  $\delta^{56}\text{Fe}$  and  $\delta^{66}\text{Zn}$  increase with decreasing MgO content in Kilauea Iki lavas, implying Fe and Zn isotopic fractionation introduced by crystal fractionation/accumulation [5, 6].

Here we present high-precision mass-dependent Ca isotopic analyses on nine basalts from Kilauea Iki lava lake using a <sup>43</sup>Ca–<sup>48</sup>Ca double spike technique. Although these samples have undergone clinopyroxene and plagioclase differentiation (up to about 20 wt% for each phase), they show no resolvable variations in Ca isotope composition, with an average  $\delta^{44/40}\text{Ca}_{\text{SRM915a}}$  of  $0.91 \pm 0.09$  (2 $\sigma$ ).

[1] Huang *et al* (2010) *EPSL* **292**, 337-344. [2] Helz (1987) *Magmatic Processes: Physicochemical Principles*. pp. 241-258. [3] Tomascak *et al* (1999) *GCA* **63**, 907-910. [4] Teng *et al* (2007) *EPSL* **261**, 84-92. [5] Teng *et al* (2008) *Science* **320**, 1620-1622. [6] Chen *et al* (2013) *EPSL* **369–370**, 34-42.