

Mesoproterozoic climate – equability in the absence of high CO₂

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Recent reconstructions based on microfossils [1], carbon isotopes [2], and paleosols [3] all indicate very low (<10x pre-industrial levels) atmospheric pCO₂ in the late Mesoproterozoic, which contrasts strongly with reconstructions from the Archean and Paleoproterozoic [4-5] that indicated pCO₂ >25x pre-industrial levels. Thus, an important question is whether the relatively low atmospheric pCO₂ indicated by proxy data is sufficient to result in the equable climate indicated by the rock record. Recent climate model results [6] indicate that even at pre-industrial pCO₂, a “hard snowball” scenario (i.e., low-latitude ice) is difficult to generate in the Neoproterozoic (Marinoan) if there is significant tropical land in the NCAR CAM3.1 model. There are two fundamental differences between that previously modeled scenario and the late Mesoproterozoic CO₂ minimum at ~1.1 Ga: 1) total solar irradiance was even lower (by ~3%), 2) there is no geologic evidence for significant ice at any latitude at that time. To investigate whether equable (and potentially ice-free) conditions are possible during the late Mesoproterozoic under a low CO₂ regime we have undertaken a series of sensitivity analyses using NCAR’s CAM3.1 model coupled with a 50 m slab ocean, variable continental land mass (including land-free), and uniform low topography. Those preliminary results indicate: 1) CH₄ was likely also important in maintaining equability during the Mesoproterozoic, which has implications for atmospheric pO₂, 2) paleogeography is an important control on water vapor and surface albedo feedbacks; failure to consider it produces biased estimates of paleoclimate sensitivity, and 3) equable (but not warm) conditions are possible at low pCO₂. Ongoing work using CESM1.2 will add a fully coupled ocean, examine variable CO₂:CH₄ mixing ratios, and explore the role of adding in paleotopography.

[1] Kah & Riding (2007) *Geology* **35** 799–802; [2] Kaufman & Xiao (2003) *Nature* **425** 279–282; [3] Sheldon (2013) *Chem. Geo.* **362** 224–231; [4] Driese *et al* (2011) *Precam. Res.* **189**:1–17; [5] Sheldon (2006) *Precam. Res.* **147**:148–155; [6] Fiorella & Poulsen (2013) *J. Clim.* **26** 9677–9695