

## High-pressure behavior of synthetic $\text{Na}_{0.884}\text{Fe}^{2+}_{0.199}\text{Mg}_{0.475}\text{Si}_{2.442}\text{O}_6$ clinopyroxene

ANDREY V. BOBROV<sup>1\*</sup>, MAXIM BYKOV<sup>2</sup>,  
 ELENA BYKOVA<sup>2</sup>, LEONID DUBROVINSKY<sup>2</sup>  
 AND LUCA BINDI<sup>3</sup>

<sup>1</sup>Geological Faculty, Moscow State University, Moscow,  
 Russia (\*correspondence: archi@geol.msu.ru)

<sup>2</sup>Bayerisches Geoinstitut, Universität Bayreuth, Bayreuth,  
 Germany

<sup>3</sup>Dipartimento di Scienze della Terra, Università di Firenze,  
 Firenze, Italy

Most clinopyroxenes of the Earth's crust and upper mantle contain only tetrahedrally coordinated silicon. However, some high-pressure experiments [1-3] demonstrate the capability of clinopyroxenes to accommodate <sup>VI</sup>Si. This suggests that the stability field of such minerals may be expanded to the lowermost upper mantle and transition zone. The first synthesis of the  $\text{Na}(\text{Mg}_{0.5}\text{Si}_{0.5})\text{Si}_2\text{O}_6$  pyroxene at 10 and 15 GPa and its unit-cell parameters measured by X-ray single-crystal diffraction were reported in [1]. The stability field of Na-pyroxene and the pyroxene/garnet transition was studied in [4]. Here we report the high-pressure behaviour of a Fe-bearing Na-pyroxene,  $\text{Na}_{0.884}\text{Fe}^{2+}_{0.199}\text{Mg}_{0.475}\text{Si}_{2.442}\text{O}_6$ , up to 30 GPa at room temperature by synchrotron X-ray single-crystal diffraction.

The diffraction pattern of a crystal, loaded in a diamond anvil cell at ESRF (ID09A), can be indexed with the following monoclinic unit cell (space group *C2/c*):  $a = 9.5059(9)$ ,  $b = 8.6667(2)$ ,  $c = 5.2715(7)$  Å,  $\beta = 108.04(9)^\circ$ ,  $V = 413.1(6)$  Å<sup>3</sup>. The lattice parameters decrease gradually with pressure reaching  $a = 9.0339(14)$ ,  $b = 8.056(17)$ ,  $c = 4.9585(7)$  Å,  $\beta = 105.47(7)^\circ$ ,  $V = 347.9(4)$  Å<sup>3</sup> at 30 GPa. No amorphization or phase transitions were detected in the studied pressure range.

The variation of the unit-cell volume is rather smooth and can be described with a third-order Birch-Murnaghan equation of state giving a bulk modulus of  $K_0 = 124(4)$  GPa and  $K' = 3.7(3)$  in the same range.

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[1] Angel *et al* (1988) *Nature* **335**, 156–158 [2] Gasparik (1989) *Contributions to Mineralogy and Petrology* **102**, 156–158 [3] Yang *et al* (2009) *American Mineralogist* **99**, 942–949 [4] Dymshits *et al* (2013) *Geochimica et Cosmochimica Acta* **105**, 1–13