Triple oxygen isotopes in terrestrial carbonates

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The ¹⁷O-anomaly (Δ^{17} O) of natural waters has been shown to be sensitive to evaporation in a manner analogous deuterium excess [1, 2]. Specifically, evaporated bodies of water (e.g., soil waters, lake waters, animal body water) will tend to have lower Δ^{17} O than primary meteoric waters. Carbonates should record the triple oxygen isotope compositions of parent waters, providing a basis in the sedimentary record for identifying evaporated waters and perhaps for estimating primary (unevaporated) δ^{18} O values of ancient meteoric water.

The Δ^{17} O of animal body water relates to factors such as intake of evaporated waters (e.g., leaf water), evaporative water effluxes, and the triple oxygen isotope composition of atmospheric O₂, which itself relates to global carbon cycling, atmospheric CO₂ levels, and stratospheric photochemistry [3, 4].

Thus there is much potential for Δ^{17} O in studies of continental paleoenvironments, but progress has been hampered by a lack of high-precision analytical methods for carbonate. We optimized a reduction / fluorination approach [5] and present the first high-precision (±0.01‰, 1σ) Δ^{17} O dataset for sedimentary and biogenic carbonates. The clearest pattern to emerge is a strong ¹⁷O-depletion in avian, dinosaurian, and mammalian body water (from analyses of eggshell and enamel) relative to meteoric waters, following expected influences of evaporated water and atmospheric O₂ on vertebrate body water. Parent waters of soil carbonates have similar or slightly lower Δ^{17} O than global precipitation, suggesting that they are mildly to moderately evaporated.

Our results suggest that Δ^{17} O will have useful application in continental environments where the effects of evaporation are important, and where animal body water may record an isotopic signal of evaporated water and atmospheric oxygen.

[1] Landais et al (2006) GCA 70, 4105-4115. [2] Luz and Barkan (2010) GCA 74, 6276-6286. [3] Pack et al (2013) GCA 102, 306-317. [4] Bao et al (2008) Nature 453, 504-506. [5] Brenninkmeijer and Röckmann (1998) Rapid. Commun. Mass Spectrom. 12, 479-483.