

Quantitative analysis of OH⁻ in fluorapatite by micro-FTIR: Determination of the molar absorption coefficient

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Introduction: Apatite [Ca₅(PO₄)₃(F,Cl,OH)] is a common mineral in igneous rocks and the only ubiquitous hydroxyl (OH⁻) bearing mineral found in planetary materials [e.g., 1-3]. Consequently, apatite is an important resource for understanding the water budgets of terrestrial and planetary magmas. Fourier Transform Infrared Spectroscopy (FTIR) offers an accurate, quick, and non-destructive method for quantifying the OH⁻ content of apatites. Quantitative measurements of molecules from the infrared spectrum are obtained by the Beer-Lambert law, $A_i = ct\varepsilon_i$, where ε_i is the integrated molar absorption coefficient (ε_i), which must be empirically determined for each molecule/mineral. This work aims to measure this value for OH⁻ in apatite.

Methods: The H₂O contents of four single-crystal, gem-quality, fluorine-rich apatites were accurately measured by continuous flow mass spectrometry (CFMS). Infrared spectra were collected with a Nicolet Nexus 670 FTIR using a microscope attachment, MCT-A detector, XT-KBr beamsplitter, and Globar source. Samples were cut perpendicular to the c-axis, doubly polished to an average thickness of 60 μm , and analysed under polarized light following procedures in [4]. Peak areas were determined using OMNIC spectral analysis software, and sample thicknesses were measured using methods outlined in [5].

Results: The calculations produced $\varepsilon_i = 2.52 \times 10^4 \pm 0.26$ L/mol/cm². This value is similar to one reported in a previous study ($2.31 \pm 0.06 \times 10^4$) [6]; however, our reported uncertainty is higher. Our uncertainty is likely more realistic because it is based on the standard deviation of ε_i calculated from the analysis of four different apatites. It is unclear how [6] computed ε_i and associated uncertainty, but treating their data set the same way as our own, we compute $\varepsilon_i = 2.02 \times 10^4 \pm 0.46$ L/mol/cm². Consequently, additional work is needed to better constrain ε_i for quantification of OH⁻ in apatite by polarized micro-FTIR.

[1] McCubbin *et al* (2010) *PNAS* **107**, 11223-11228 [2] McCubbin *et al* (2012) *Geology* **40**, 683-686 [3] Sarafian *et al* (2013) *Met. Planet. Sci.* **48**, 2135-2154 [4] Libowitzky & Rossman (1996) *Phys. Chem. Min.* **23**, 319-327 [5] Berger (2012) M.S. Thesis, UNM, 71pp. [6] Wang *et al* (2011) *Amer. Min.* **96**, 1392-139.