

ATR-FTIR spectroscopy to measure H₂O and CO₂ in silicate glass

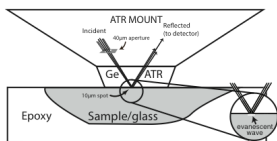
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Over the past two years, we have developed protocols to measure H₂O and CO₂ in silicate glass with Attenuated Total Reflectance (ATR) FTIR spectroscopy. Our initial calibration [1] demonstrated the feasibility of the technique for quantifying total H₂O (H₂O_t), both as molecular H₂O (H₂O_m) and OH in glass within polished epoxy mounts prepared for EPMA and SIMS analysis.

In our setup, an ATR accessory crystal (Ge) is inserted into the FTIR microscope, the IR source beam is directed through the crystal, and a background spectrum is collected in air. By raising the stage, the sample makes contact with the Ge crystal, such that a small percentage of the incident light is absorbed within the upper few hundred nm of the sample. Our initial study [1] demonstrated that absorbance at 3450 cm⁻¹ and 1630 cm⁻¹ correlated with H₂O_t and H₂O_m concentration, respectively. This permits quantification of these species in silicate glass (calc-alkaline basalt through rhyolite) with a single calibration, adjusted for glass density. Other ATR crystal types and geometries will likely require separate calibration curves.

Recently, we explored calibrations for dissolved carbonate in basaltic and basaltic andesite glasses, as well as molecular CO₂ in rhyolitic glass. In both cases, calibration is clearly feasible, though with detection limits significantly higher than for transmission FTIR. By using a N₂-purge in our FTIR microscope to minimize atmospheric CO₂, we can resolve dissolved molecular CO₂ as low as 150 ppm. Carbonate peaks at 1400-1500 cm⁻¹ are not affected by atmospheric CO₂, though H₂O vapor can interfere with resolution of the peaks, and as with transmission FTIR, peak heights are difficult to define due to the presence of the sample surface. Additionally, the detection limits are affected by the refractive index of the sample. Our future work will focus on: 1) improving the detection limits for CO₂ in silicate glass, 2) constraining the variation in the refractive index coefficient) with glass composition.



[1] Lowenstern & Pitcher (2013) *Am. Mineral.* **98**, 1660-1668.