

Vaporization Studies of Olivine via Knudsen Effusion Mass Spectrometry

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Vaporization and condensation are important in a variety of cosmochemical processes, including contributions to the atmospheres of the hot, rocky exoplanets. Constituents of these atmospheres have been estimated from vaporization calculations for various planetary analogs [1], such as the continental crust, basic silicate earth, and moon composition. Experimentally, Knudsen Effusion Mass Spectrometry (KEMS) has been effectively utilized to analysis the vapor species above similar compositions [2-3].

A KEMS instrument is based on a Knudsen cell, which is a small enclosure where near solid/vapor equilibria is attained. The enclosure contains a well-defined orifice which forms a molecular beam of the vapor. This molecular beam is then directed into a mass spectrometer for analysis. This study is conducted on two mass spectrometers—a fast scanning quadrupole instrument for determination of the vapor composition and a magnetic instrument for more detailed analysis of the pressures of each species [4].

KEMS is primarily an equilibrium technique. However, kinetics are well-known to play a role in natural vaporization processes, particularly with oxides. These are quantified with the vaporization coefficient, which is defined as the ratio of the flux leaving a surface to the equilibrium flux. For an oxide, vaporization coefficients vary from ~ 0.5 to 10^{-3} [5]. It is shown how these can be measured by varying the orifice geometry in a multi-cell KEMS experiment.

Initial results for a fosterite-rich olivine ($\text{Fo}_{93}\text{Fa}_7$) will be presented and compared to theoretical predictions.

[1] L. Schaffer and B. Fegley, Jr.(2009), *Astrophys. J.* **703**, L113 [2] V. Piacente, *et al* (1975), *Silikaty C.* **4**, 289-297 [3] O. M. Markova, *et al* (1986), *Geokhimiya* **11**, 1559-1568 [4] E. H. Copland and N. S. Jacobson (2010), NASA TP-2010-216795 [5] A. W. Searcy *et al* (1970), *Chemical and Mechanical Behavior of Inorganic Materials*, Wiley, p. 128