

Mineral characteristics of Tungsten-bearing granite in the Jiangnan orogenic belt: A case study of the Qingyang pluton

JIE ZHOU^{1,2*}, YAO-HUI JIANG¹ AND WEIYA, GE²

¹ State Key Laboratory for Mineral Deposits Research, School of Earth Sciences and Engineering, Nanjing University, Jiangsu, 210093, P.R. China (*correspondence: zhoujie0517@sina.com)

² Nanjing Institute of Geology and Mineral Resources, China, 210016, P.R.China

South China is the most important tungsten-concentrated zone in China and even in the world, famous for the Nanling Tungsten ore belt. A series of Tungsten deposits, related with the late Yangshanian granites, have been recently discovered in the Jiangnan orogen, northeastern Qin Zhou-Hangzhou metallogenic belt. The Qingyang composite granite is one of a tungsten-bearing granites in the east of Jiangnan orogenic belt. It crops out over an area of approximately 750 km². The Baizhangya tungsten-molybdenum deposit was found near this pluton. New LA-ICPMS zircon U-Pb dating suggests that the crystallization age of the Qingyang body is 145.5±0.5Ma. The Qingyang pluton has zircon, apatite, fluorite, titanite, rutile, ilmeite, limonite, anatase and magnetite. The main rock-forming minerals including amphibole, biotite, plagioclase were analysed chemical compositions. The biotite is characterized by high MgO and low FeO contents with high Mg/(Mg+Fe) ratios (0.58-0.61), plotted in the crust-mantle zone. The oxygen fugacity calculated by biotite compositions is above Ni-NiO (NNO). The amphiboles are magnesiohornblende. The pressure of the granite estimated by Al-in-hornblende barometer is 1.79~2.50kbar. An amphibole-plagioclase thermometry and a semiquantitative hornblende thermometer yield a forming temperature of ~714°C. The plagioclase is oligoclase (An~29.5%). The mineral characteristics of Qingyang pluton are different from the Nanling tungsten-bearing S-type granites. The Jiangnan tungsten ore belt is NE-trending and distinct from the Nanling belt in terms of metallogenic age, tungsten-bearing granite type etc. and require further study.

Molecular simulation study of rectorite

ZHOU JINHONG^{1,2*}, LU XIANCAI³, ZHU JIANXI¹, MICHIEL SPRIK⁴, EDO BOEK AND HE HONGPING¹

¹ Key Laboratory of Mineralogy and Metallogeny, Guangzhou Institute of Geochemistry, Chinese Academy of Sciences, Guangzhou 510640, P.R. China

² Graduate University of Chinese Academy of Sciences, Beijing 100049, P.R. China

³ State Key Laboratory for Mineral Deposits Research, School of Earth Sciences and Engineering, Nanjing University, Nanjing 210093, P.R. China

⁴ Department of Chemistry, University of Cambridge, Cambridge CB2 1EW, United Kingdom

⁵ Department of Chemical Engineering, Imperial College London, London SW7 2AZ, United Kingdom

Rectorite is a special kind of clay mineral, consisting of illite layer and smectite layer in a regular order [1]. Rectorite and organo-intercalated rectorite can be applied in many fields. In this study, we use molecular simulations to investigate the interlayer properties of pristine and organo-rectorites. First, we use grand canonical monte carlo and molecular dynamics methods to investigate the hydration properties of rectorite with the comparison with montmorillonite. The results indicate that rectorite shows a similar swelling pattern as montmorillonite but a different interlayer cation distribution [2]. Second, we employ classical molecular dynamic simulations to study the microscopic interlayer properties of HDTMA⁺-intercalated rectorites with and without water at different HDTMA⁺ loading levels [4]. According to our simulations of organo rectorite, we find that as the loading level changes, different configurations of HDTMA⁺ occur. And water addition leads little influence on the mobility of Na⁺, but decreases the mobility of alkyl chains. Also we observe the behaviour of anions in system exceeding 1 CEC.

[1] Weaver (1956) *American Mineralogist* **41**, 202-231. [2] Kawano & Tomita (1992) *Clays and Clay Minerals* **40**, 421-428. [3] Zhou & Lu (2012) *Journal of Physical Chemistry C* **116**, 13071–13078. [4] Cygan & Greathouse & Heinz & Kalinichev (2009) *Journal of Materials Chemistry* **19**, 2470-2481.