Hydrologic variability in coastal southwest USA

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Understanding hydroclimate variability is a pressing need in the world's drylands, yet it is here that climate projections differ in the sign and magnitude of change in precipitation [1]. Within the southwest USA a different precipitation regime influences the California coast (winter only) and the interior (bimodal winter and summer). Here, we present multi-proxy reconstructions of the hydroclimate of the coastal zone. Sediments from Zaca Lake capture a sub-decadal history of wet-dry oscillations spanning 3ka whereas Lake Elsinore records climatic transitions from 33ka to Holocene.

Hydrogen isotopes in plant leaf waxes provide a record of the changing hydrological regime. Over the last 3ka, δD values of the C_{28} n-alkanoic acid are highly variable across -175% to -125% in Zaca Lake. Grain size records multidecadal variations in runoff and charcoal reveals the variable history of fire. At Lake Elsinore, a longer term shift from δD values of -190% at 19ka, to -120% at 9ka, is attributed to a shift from North Pacific to sub-tropical Pacific moisture sources, paralleling the radiative forcing trend out of the last glacial. Similarly, grain size records a long-term decrease in runoff and pollen record a transition from *Pinus* to *Quercus* indicating warming and drying. Southwest climate differs during the Younger Dryas; wet conditions found inland are linked to AMOC forcing [2]; here we show the coast remains dry.

[1] Solomon et al., (2009) PNAS, **106**, 1704-1709. [2] Clark et al., (2012) PNAS, **109**, E1134-1142.

Diamond dissolution in COH fluids

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Morphology of surface features on natural kimberliteborne diamonds indicates their interaction with fluid and greatly depends on the composition of this diamonddestructing fluid. H₂O:CO₂ ratio in the fluid dictates the degree of rounding, intensity of the pit development on {111} face, orientation and morphology of trigonal pits, and presence of other less common etch features (e.g. circular pits, hexagons). A number of different mechanisms have being proposed to explain the diversity of diamond resorption features. Application of experimental findings to natural diamonds demonstrated a relationship between diamond morphology and behavior of volatiles in kimberlite magma. However, absence of a model that relates diamond micromorphology to the characteristics of the reacting fluid precludes more quantitative characterization of kimberlite volatile system using diamond micro-morphology. Toward this end, we conducted atomic force microscopy (AFM) study of diamond crystals from dissolution experiments with H₂O and CO₂-fluid and from natural kimberlites.

Diamond octahedra were etched in H₂O or CO₂ fluid at 1 GPa and temperature (T) = 1150-1350°C. AFM data collected in tapping and contact mode demonstrated that outside of few large etch pits (up to 2 um depth) the roughness of the diamond surface is less than 400 nm in H₂O fluid and increases with T. In CO₂ fluid the roughness is between 600nm and 2 um and is independent on T. Trigonal etch pits in H₂O fluid have constant diameter independent on the depth but increasing with T. In CO₂ runs trigonal pits show continuos sizes and positive correlation between the diameter and the depth. In H₂O fluid trigons are mostly flat-bottomed, whereas in CO2 fluid they develop round bottom at higher T and more pointy-bottom at lower T. The general resorption morphology and presence of hexagonal pits in CO2-bearing runs could be a result of the different dissolution rate in [111], [100], and [110] direction in fluids with various H₂O:CO₂ ratio. However, the more regular layer-by-layer resorption in H2O fluid and irregular deeper material removal in CO2 fluid may also indicate different mechanism of interaction of surface complexes formed in CO2- and H2O-dominated media and different rate of removal of 3- and 2-bonded atoms from diamond surface by these complexes. We conducted AFM measurments of trigonal pits on 22 micro-diamonds from four Canadian kimberlite pipes, which different geology indicates different behavior of volatiles. We use this data to examine magmatic fluid and the relative crystallization T.