

Methane geochemistry's 'stealth' process: Microbial oxidation

WILLIAM S. REEBURGH

Earth System Science, University of California Irvine, Irvine, CA, 92697, USA (reeburgh@uci.edu)

For the purposes of this presentation, I consider a 'stealth' process as one that cannot be directly measured. Microbial methane oxidation, aerobic and anaerobic, are examples. Most of the fluxes used to assemble the global methane budget are net fluxes, the difference between production and consumption, so direct measurements of production and consumption require tracer measurements. Several recent high-profile climate modeling papers have fallen into the 'stealth process' trap by failing to explicitly consider microbial oxidation.

This talk reviews recent developments in quantifying methane oxidation in the oxic ocean and estimates the global importance of both aerobic and anaerobic oxidation in a range of environments as controls on global change induced production. New methods are discussed and remaining challenges are enumerated.

Phase separation, degassing and anomalous methane at the Menez Gwen hydrothermal field

EOGHAN P. REEVES¹, X. PRIETO¹, M. HENTSCHER¹,
M. ROSNER², J. SEEWALD³, K.-U. HINRICHS¹
AND W. BACH¹

¹MARUM Center for Marine Environmental Sciences,
Universität Bremen, Bremen 28359, Germany
(*correspondence: reeves@uni-bremen.de)

²Fachbereich Geowissenschaften, Freie Universität Berlin,
Berlin 12249, Germany

³Woods Hole Oceanographic Institution, Woods Hole MA
02543, USA

The Menez Gwen neovolcanic dome, lying at ~800m depth, hosts one of the shallowest basalt-hosted hydrothermal systems on the Mid-Atlantic Ridge and vent fluids there have unusually high CH₄/H₂ ratios relative to ultramafic- and other basalt-hosted systems [1]. To further constrain the origin of this CH₄, fluids were sampled in 2010 from vents on the dome's eastern flank, and from a newly discovered vent field, Bubbylon, 5km to the south of the Menez Gwen dome.

Maximum vent temperatures (270–298°C) in the eastern Hot Sands and AzorAna areas are near the 2-phase boundary and near-seafloor subcritical phase separation (boiling) is pervasive. Endmember CH₄ (0.37–3.6mmol/kg) and CO₂ (11–78mmol/kg) concentrations are high, but H₂ (0.06–0.57mmol/kg) is low, and only a narrow Cl range (202–368mmol/kg) is evident. Two volatile-rich diffuse fluids (91–117°C) sampled appear to have formed by subsurface mixing of near zero salinity vapor phases with entrained seawater, with apparent CH₄ and CO₂ end-member concentrations of ~30 and ~600mmol/kg, respectively. Such extreme volatile enrichments due to boiling are likely responsible for the ubiquitous gas bubble exsolution evident at the seafloor, releasing substantial CO₂ and CH₄ to the water column. Despite boiling-related variability, endmember CH₄/H₂ ratios (2.5–18) remain anomalously high in 2010. Trace NH₄⁺ concentrations (<5µmol/kg) and δ¹³C_{CH4} values (-16.6‰ to -19.9‰) within the typical range of unsedimented systems preclude any subsurface sedimentary source of thermogenic CH₄ [2]. Thermodynamic considerations also exclude abiotic CO₂ reduction to CH₄ at the conditions encountered by fluids. This suggests a deeper source of CH₄, such as the volatile-rich fluid inclusions found in oceanic crust Layer 3 [3]. CH₄ radiocarbon and C₂₊ hydrocarbon analyses are ongoing.

[1] Charlou *et al.* (2000) *Chem. Geol.* **171**, 49–75. [2] Lilley *et al.* (1993) *Nature* **364**, 45–47. [3] Kelley & Früh-Green (1999) *J. Geophys. Res.* **104**, 10439–10460.