

## Derivation of a new soil respired CO<sub>2</sub> proxy for application to paleo-CO<sub>2</sub> reconstruction

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One way to predict future climate change is by studying warming events in the geologic past. The main control of the Earth's temperature over geologic time is the concentration of atmospheric CO<sub>2</sub>; however, the precise amount of CO<sub>2</sub> in the past is not fully understood [1]. Therefore, it is essential to quantify the atmospheric CO<sub>2</sub> concentration in the geologic past, especially at times of rapid climate change analogous to current changes.

The widely applied paleobarometer introduced by Cerling [2] calculates atmospheric CO<sub>2</sub> based on the isotopic ratio of pedogenic carbonates precipitating in equilibrium with soil CO<sub>2</sub>. Its primary source of uncertainty is the amount of soil respired CO<sub>2</sub> in the paleosol at the time of pedogenic carbonate formation. To remove the uncertainty from the Cerling [2] paleobarometer, a new soil respired CO<sub>2</sub> proxy was calibrated for soils and paleosols. The relationship between modern soil CO<sub>2</sub> and mean annual precipitation was derived from an extensive literature review of modern carbonate bearing soil CO<sub>2</sub> data. Based on 580 measurements of summer soil CO<sub>2</sub> (time of carbonate formation) from 21 pedogenic carbonate-bearing soils, we find a linear relationship between summer levels of soil respired CO<sub>2</sub> and MAP ( $r^2 = 0.63$ ; SE = 1126 ppm). For paleosols, past soil respired CO<sub>2</sub> can be predicted using MAP estimates calculated from either the depth to Bk horizon [3] or the chemical index of alteration without potassium [4].

During the Middle Miocene Climatic Optimum (16Ma) we see elevated atmospheric CO<sub>2</sub> levels analogous to the future rise in CO<sub>2</sub> over the next century [5]. Here we use this new soil respired CO<sub>2</sub> proxy to reconstruct atmospheric CO<sub>2</sub> during the middle Miocene from Montana and compare these results to previously published estimates from Montana and Kenya during this time period [5], [6].

[1] IPCC (2007) *Synthesis Report*, 104 pp. [2] Cerling (1991) *American Journal of Science* **291**, 377–400. [3] Retallack (2005) *Geology* **33**, 333–336. [4] Sheldon, Retallack & Takana (2002) *Journal of Geology* **110**, 687–696. [5] Retallack (2009) *Palaeogeography, Palaeoclimatology, Palaeoecology* **281**, 57–65. [6] Ekart, Cerling, Montanez & Tabor (1999) *American Journal of Science* **299**, 805–827.

## Basaltic glasses as records of mantle oxygen fugacity

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Several petrological lines of evidence indicate that basalts from different tectonic environments (MORB, BABB, OIB, arcs) express different inherent oxygen fugacities. It is ambiguous whether  $fO_2$ -diversity derives from the mantle source or shallow processing. Fe<sup>+3</sup>/ΣFe ratios of primitive, undegassed basaltic glasses and naturally glassy olivine-hosted melt inclusions, determined using μ-XANES [1], may reflect the oxygen fugacity of the mantle source. Major, trace, and volatile elements, and isotopic ratios measured on the same glasses, allows us to interrogate the influences on mantle and magmatic redox.

Fe<sup>+3</sup>/ΣFe ratios of MORB glasses are  $0.16 \pm 0.01$  (n=60), higher than reported by micro-colorimetry ( $0.07 \pm 0.03$ ) [2] or direct titration ( $0.12 \pm 0.02$ ) [3]. Only 0.01 (absolute) of the XANES/wet-chemistry disparity can be attributed to our Mossbauer-based XANES calibration, and may derive from differences in the material being sampled and the influence of FeO\* uncertainty on wet-chemical determinations. MORB glasses record olivine control of Fe<sup>+3</sup>/ΣFe ratios. Primitive glasses, with only olivine on the liquidus, are traced back to primary mantle melts by incrementally adding equilibrium olivine until equilibrium with Fo<sub>90</sub> is reached, and then referenced to  $fO_2$  [4]. Primary MORB magmas indicate  $fO_2$  of QFM  $-0.25 \pm 0.18$  (referenced at 1200°C, 1 atm).

Calculated primary melts for the most primitive arc melt inclusions from the Marianas reflect QFM  $+0.84 \pm 0.22$  while BABB varies between arc and MORB sources as a function of primary melt water content – mirroring the correlation seen in the raw Fe<sup>+3</sup>/ΣFe ratios [5]. We show that degassing and crystallization cannot be responsible for these observations. The influence of the oxidized plate appears to persist through plate recycling. Samoan lavas are more oxidized than average MORB, and their Fe<sup>+3</sup>/ΣFe ratios correlate with EM2-type isotopic enrichment that can be attributed to hydrated, oxidized, subducted sediment.

[1] Cottrell *et al.* (2009) *Chem. Geol.* **268**, 167–169. [2] Christie *et al.* (1986) *EPSL* **79**, 397–411. [3] Bezos & Humler (2005) *EPSL* **69**, 711–725. [4] Kress & Carmichael (1991) *CMP* **108**, 82–92. [5] Kelley & Cottrell (2009) *Science* **325**, 605–607.