## Lithospheric mantle beneath East Antarctic (Oasis Jetty)

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Some mantle-derived East Antarctic rocks have bulk-rock and mineral trace element and isotopic compositions that provide evidence of strong carbonate metasomatism. 46 peridotite xenoliths from lithospheric mantle beneath Lambert-Amery Rif were studied. The age of hosting ultrabasic alkaline dykes is cretaceous. Investigated xenoliths represents about 80 km thick cross of subcontinental litospheric mantle.

Distinct correlations is observed between contents of MgO and other petrogenetic elements.Comparison of these correlations with estimated trends of batch and fractional melting of initially fertile mantle peridotite [1] demonstrate, that Jetty peridotites plot at lower values for SiO<sub>2</sub>. Garnets are characterized by unusual high LIL element concentrations (Ba/La=4000) and clinopyroxenes from the spinel peridotite suite are highly enriched in LREE [(La/Yb)<sub>N</sub> =14] and in HFSE .The decoupling of major and REE elements and observed mineral assemblages ( which include dolomite, calcite, apatite, Ti- Ba-rich phlogopite, titanite, perovskite, henrymeierite) indicate that the upper mantle of East Antarctic has been affected by very strong metasomatism. Henrymeierite was found in mantle material for the first time [2].Our isotopic data showed complete absence of isotope equilibrium among the rock forming minerals in the xenoliths.Garnets are characterised by the most radiogenic Sr isotope composition (0.709024-0.714127), whereas Cpx exhibits the lowest <sup>87</sup>Sr/<sup>86</sup>Sr ratios ( 0.702861-0.703711). Garnets are depleted in terms of Sm-Nd system and they are significantly enriched in radiogenic Sr. Such anomalous signature in Antarctic mantle rocks confirm the influx of carbonate metasomatising fluid ( or melt) from the source with a time integrated LIL-elements enrichment and more rapid Sr but not Nd diffusion to garnets compare to Cpx. It is probable, that that the interaction of primary dolomitic melts-fluids with mantle material according to reactions:  $2Mg_2Si_2O_6$  +  $CaMg(CO_3)_2 = 2Mg_2SiO_4 + CaMgSi_2O_6 + 2CO_2$ (1)  $3CaMg(CO_3)_2 + CaMgSi_2O_6 = 4CaCO_3 + 2Mg_2SiO_4 + CO_2$  (2) resulted in wehrlitization and carbonatization. The fluids responsible for this metasomatism were probably derived from the plume that arrived beneath the region at this time. References

[1] Niu Y. (1997) *J.Petrology*, **38**, 1047-1074. [2] Kogarko L., Kurat G., Ntaflos T. (2007) Can *Min*, **45**, 497-501

## Identification of distinct sources of water and solutes from temporal variations of chemical and isotopic compositions during a pumping test in a test borehole

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Chemical and isotopic compositions of groundwater were measured during a 70-hr pumping test for a test borehole drilled in volcanic aquifers consisting of trachytic rocks, scoria and hydrovolcanic tuffs, where natural land cover is dominant. Groundwater from initial sampling using a bailer was Na(Ca)-HCO<sub>3</sub> type with low mineral contents. <sup>3</sup>H levels indicated recent recharge, which was similar to those of rainwater. However, chemical and isotopic compositions of groundwater showed significant temporal variations during the pumping test. The mineral contents steadily increased with pumping by three-fold and water types varied from Na(Ca)-HCO<sub>3</sub> to Mg-HCO<sub>3</sub> with little traces of anthropogenic contamination. Among solutes, concentrations of HCO<sub>3</sub> and Mg significantly increased and those of other cations increased in lesser extent whereas concentrations of Cl, SO<sub>4</sub>, NO<sub>3</sub> and SiO<sub>2</sub> were nearly invariant indicating increases in mineral contents were mainly attributed to water-rock interactions. However, pH decreased with mineral content increase suggesting additional CO<sub>2</sub> sources.  $\delta^{13}$ C of dissolved inorganic carbon clearly showed the sources of HCO<sub>3</sub>. Depleted  $\delta^{13}$ C (-18.6‰) was observed in the early stage of pumping showing dissolution of biological CO<sub>2</sub> from soil zones while enriched  $\delta^{13}$ C (-6.1‰) in the later stage of pumping indicating contribution of deep-seated, presumably, magmatic CO<sub>2</sub>. <sup>3</sup>H values varied from 3.0 TU to 2.1 TU during pumping, seemingly indicating water was still contributed from recent water in the later stage because highmineral water is likely to have negligible <sup>3</sup>H, which was corroborated by very low levels of CFCs. The chemical and isotopic characteristics suggested that contribution of highmineral water was significant for most solutes but, not for water in groundwater obtained from continued pumping.